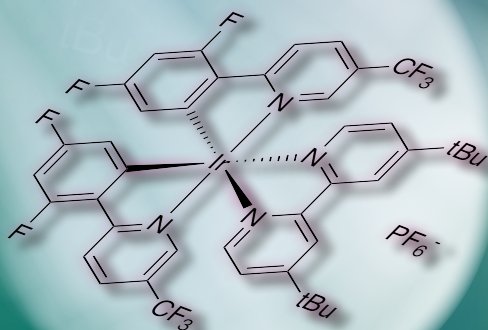


# Photocatalysts



**TREM**

# Photocatalysts

<b>Glossary of Terms</b> .....	i
<b>Article: Iridium and Ruthenium Photocatalysts for Visible Light Photocatalysis in Organic Synthesis</b> .....	1-11
<b>Quick Reference of Photocatalyst Products</b> .....	12-13
<b>Photocatalysts</b> Includes technical notes and complete product details (sorted by key element)..	14-29
<b>Photocatalyst Kits</b> .....	30-34
<b>Photochemical Equipment and Related Kits</b> .....	35-42
<b>Available Booklets</b> .....	44

Strem Chemicals has been providing fine chemicals for research and commercial production for over fifty years. In this booklet you will find our selection of photocatalysts, related kits for screening purposes, as well as ligands and precursors for photocatalyst synthesis. There have been interesting developments in photocatalysis in the past few decades. Many of these novel compounds have been applied to solar cell research, light emitting diode manufacturing (LED) and initiators for free radical polymerizations. Specifically, cyclometallated ruthenium and iridium complexes are the most prominent. Recently, these compounds have been successfully applied to catalytic transformations. These catalysts can also be utilized in challenging organic transformations in both bench-top and commercial scales.

At Strem, we also offer a wide variety of ligands, nanomaterials and CVD/ALD precursors. Most of our products are of high purity, typically at 99%, while some are as high as 99.9999% metals purity. We continually seek to provide new technologies from around the globe and add to our product line. We have licensing agreements with industry and academia, which allow easier access to these patent-protected products for our customers. We look forward to continued growth in order to best serve our customers' needs with the quality and service they can trust from Strem.

As part of our ongoing commitment to quality, we have achieved ISO 9001 certification for the Quality Management System (QMS) at our corporate headquarters in Newburyport, Massachusetts.

In addition, custom synthesis services are provided on a contract basis. For pharmaceutical applications, manufacturing is conducted under current Good Manufacturing Practices (cGMP) in FDA inspected kilo-lab suites. Complete documentation is available, including validation and stability studies. Active Drug Master Files (DMF's) are maintained in North America and Europe.

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Our other booklets, which focus on applications and product classes, are available in print per request and also on our website. Below is a list of current booklet titles that are available. Please also check our Product Resources section online to find additional literature offerings, such as the Strem Chemiker, our technical publication, and product literature sheets.

- Biocatalysts
- Buchwald Ligands and Precatalysts
- Carbon-Base Nanomaterials & Elemental Forms
- Catalysts & Ligands *Sold in Collaboration with Takasago*
- Chiral Phosphoric Acids
- Gold Elements & Compounds
- Heterogeneous Catalysts
- High Purity Chiral Reagents *Sold in Collaboration with Daicel*
- Kits
- Materials for Energy Applications
- Metal Catalysts for Organic Synthesis
- Metathesis Catalysts
- MOCVD, CVD & ALD Precursors
- MOFs and Ligands for MOF Synthesis
- Nanomaterials
- New Products
- Other Ligands
- Phosphorous Ligands and Compounds
- Photocatalysts
- PURATREM: High Purity Inorganics

Ephraim S. Honig, Ph.D., M.B.A  
Chief Executive Officer



*Photocatalyst 12/18*

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# Glossary of Terms

<b>[α]<sub>D</sub></b>	.....	Specific rotation
<b>AAS</b>	.....	Atomic Absorption Standard
<b>ACS</b>	.....	Conforms to American Chemical Society specifications
<b>air sensitive</b>	.....	Product may chemically react with atmospheric oxygen or carbon dioxide at ambient conditions. Handle and store under an inert atmosphere of nitrogen or argon.
<b>amp</b>	.....	Ampouled
<b>b.p.</b>	.....	Boiling point in °C at 760mm, unless otherwise noted
<b>d.</b>	.....	Density
<b>dec.</b>	.....	Decomposes
<b>elec. gr.</b>	.....	Electronic Grade, suitable for electronic applications
<b>f.p.</b>	.....	Flash point in °F
<b>gran.</b>	.....	Granular
<b>heat sensitive</b>	.....	Product may chemically degrade if stored for prolonged periods of time at ambient temperatures or higher. Store at 5°C or lower.
<b>hydrate</b>	.....	Unspecified water content which may vary slightly from lot to lot
<b>hygroscopic</b>	.....	Product may absorb water if exposed to the atmosphere for prolonged periods of time (dependent on humidity and temperature). Handle and store under an inert atmosphere of nitrogen or argon.
<b>light sensitive</b>	.....	Product may chemically degrade if exposed to light
<b>liq.</b>	.....	Liquid
<b>m.p.</b>	.....	Melting point in °C
<b>moisture sensitive</b>	.....	Product may chemically react with water. Handle and store under an inert atmosphere of nitrogen or argon.
<b>NMR grade</b>	.....	Suitable as a Nuclear Magnetic Resonance reference standard
<b>optical grade</b>	.....	For optical applications
<b>pwdr.</b>	.....	Powder
<b>primary standard</b>	.....	Used to prepare reference standards and standardize volumetric solutions
<b>PURATREM</b>	.....	Product has a minimum purity of 99.99% (metals basis)
<b>purified</b>	.....	A grade higher than technical, often used where there are no official standards
<b>P. Vol.</b>	.....	Pore volume
<b>pyrophoric reagent</b>	.....	Product may spontaneously ignite if exposed to air at ambient conditions
	.....	High purity material, generally used in the laboratory for detecting, measuring, examining or analyzing other substances
<b>REO</b>	.....	Rare Earth Oxides. Purity of a specific rare-earth metal expressed as a percentage of total rare-earths oxides.
<b>SA</b>	.....	Surface area
<b>store cold</b>	.....	Product should be stored at -18°C or 4°C, unless otherwise noted (see product details)
<b>subl.</b>	.....	Sublimes
<b>superconductor grade</b>	.....	A high purity, analyzed grade, suitable for preparing superconductors
<b>tech. gr.</b>	.....	Technical grade for general industrial use
<b>TLC</b>	.....	Suitable for Thin Layer Chromatography
<b>v.p.</b>	.....	Vapor pressure mm of Hg
<b>xtl.</b>	.....	Crystalline

## About Purity

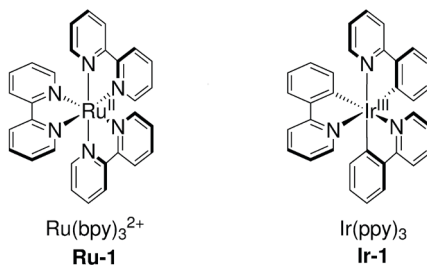
<b>Chemical purity</b>	.....	is reported after the chemical name, e.g. Ruthenium carbonyl, 99%
<b>Metals purity</b>	.....	is reported in parentheses with the respective element, e.g. Gallium (III) bromide, anhydrous, granular (99.999%-Ga) PURATREM where 100% minus the metal purity is equal to the maximum allowable percentage of trace metal impurity

# Iridium and Ruthenium Photocatalysts for Visible Light Photocatalysis in Organic Synthesis

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## Introduction

Photoredox catalysis has, in the past decade, grown to become a commonly employed catalytic manifold for the construction of molecular complexity in unique and powerful ways.<sup>1</sup> In particular, C–C and C–heteroatom bond constructions have been enabled by the intermediacy of open-shell or electronically-excited intermediates generated by single-electron transfer (SET) or energy transfer (ET). The most fruitful catalyst frameworks to emerge have been those of homoleptic ruthenium and homo- and heteroleptic iridium polypyridyl complexes, of the Ru(bpy)<sub>3</sub><sup>2+</sup> (**Ru-1**) and Ir(ppy)<sub>3</sub> (**Ir-1**) framework, previously used in dye-sensitized solar cells,<sup>2</sup> as emitters in phosphorescent OLEDs,<sup>3</sup> photocatalysts in water splitting<sup>4</sup> and CO<sub>2</sub> reduction,<sup>5</sup> and in oxygen sensing<sup>6</sup> (Figure 1). However, as more complex organic reactivity has been explored and developed, the use of functionalized ligands on the metal center has proven necessary.



**Figure 1.** Ru(bpy)<sub>3</sub><sup>2+</sup> and Ir(ppy)<sub>3</sub>, commonly employed photocatalysts

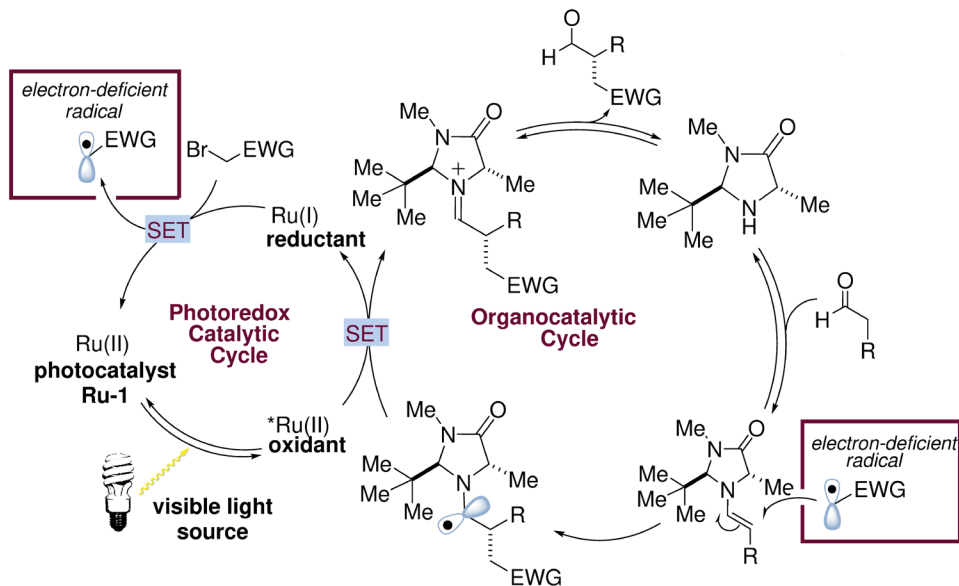
The use of specialized photocatalysts with rationally designed ligand scaffolds has become commonplace, as characteristics such as oxidizing or reducing power, excited state lifetime, and triplet excited state energy have been optimized for specific transformations or catalytic platforms. As such, robust synthetic methods for the rapid generation of differentially substituted ruthenium and iridium polypyridyl complexes have been developed, enabling a variety of synthetic transformations.

## History of Photocatalysts in Other Applications

Previous to the use of ruthenium and iridium polypyridyl complexes as photocatalysts in synthetic organic chemistry, a rich literature had been developed for their use in other applications. Their ability to perform photo-initiated electron transfer enabled their use as photosensitizers in water splitting, and subsequent work demonstrated their utility in dye-sensitized solar cells. More relevantly, limited reports appear sporadically in the literature describing the use of Ru(bpy)<sub>3</sub><sup>2+</sup> as a photocatalyst in organic transformations prior to the current era. In 1981, Pac and coworkers demonstrated a photocatalytic reduction of electron-deficient olefins via neutral  $\alpha$ -acyl radicals, using 1-benzyl-1,4-dihydroquinoline (BANH) as the terminal reductant.<sup>7</sup> Similar transformations for the reduction of activated alkyl halides via the fragmentation of neutral alkyl radicals and halide anions have also been reported.<sup>8</sup> Additionally, some Ru(bpy)<sub>3</sub><sup>2+</sup>-mediated net oxidative transformations had appeared in the literature prior to 2000.<sup>9</sup>

## Ruthenium Photocatalysts

In 2008, we published an enantioselective  $\alpha$ -alkylation of aldehydes using a combination of chiral amine organocatalysis and Ru(bpy)<sub>3</sub><sup>2+</sup> (**Ru-1**) photoredox catalysis (Scheme 1).<sup>10</sup> This transformation proceeded via initial quenching of the photocatalyst excited-state \*Ru(bpy)<sub>3</sub><sup>2+</sup> by a sacrificial amount of enamine to generate the highly reducing Ru(bpy)<sub>3</sub><sup>+</sup> (not shown). Then, single-electron transfer (SET) from this Ru<sup>I</sup> state to an alkyl bromide could induce fragmentation to afford bromide anion and a neutral electron-deficient radical. This electrophilic radical can add to a catalytically-generated enamine to forge the new C–C bond and generate an  $\alpha$ -amino radical. Then, SET oxidation of this species could be accomplished by \*Ru(bpy)<sub>3</sub><sup>2+</sup> to yield the product, after hydrolysis of the organocatalyst.

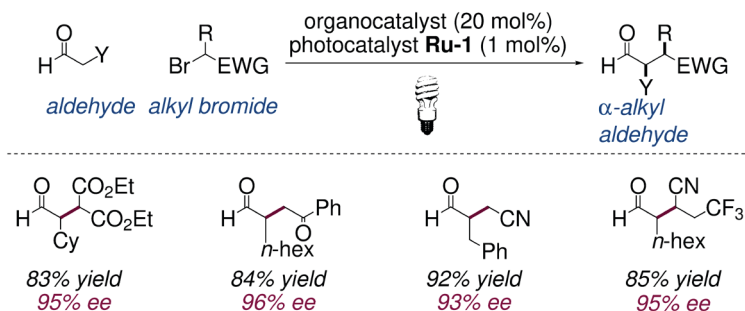


**Scheme 1.** Catalytic cycle of enantioselective  $\alpha$ -alkylation of aldehydes using **Ru-1**

As such, the ruthenium photocatalyst could perform both SET oxidation and reduction in the same reaction, enabling a redox neutral, room-temperature, light-driven radical pathway. This mechanism was also extended in 2015 to accommodate bromoacetonitrile derivatives as the alkyl radical precursor,<sup>11</sup> and a representative scope of this general reaction manifold is shown in Table 1.

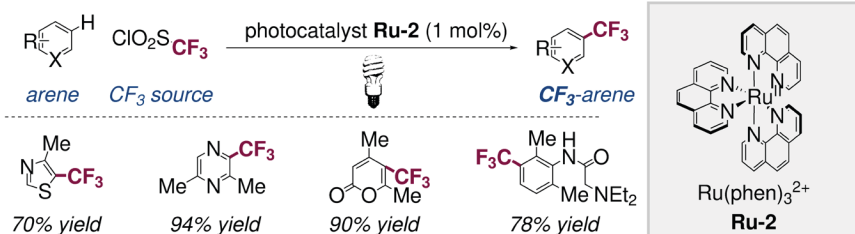
Concurrent with our publication the Yoon<sup>12</sup> group, followed shortly thereafter by the Stephenson<sup>13</sup> group, published different methodologies which similarly took advantage of the ability of the reduced Ru(bpy)<sub>3</sub><sup>+</sup> state to perform challenging single-electron reductions of organic substrates. These contemporaneous reports sparked the interest of the synthetic organic community in utilizing ruthenium photocatalysts to enable open-shell mechanistic pathways, leading to a rapid growth in the number of publications concerning synthetic organic photoredox catalysis.

Other ruthenium-based photocatalysts have also been successful for a variety of chemical transformations. Within our group, in particular, we accomplished the direct C–H trifluoromethylation of arenes with trifluoromethyl radical derived from reduction of triflyl chloride, CF<sub>3</sub>SO<sub>2</sub>Cl, mediated by



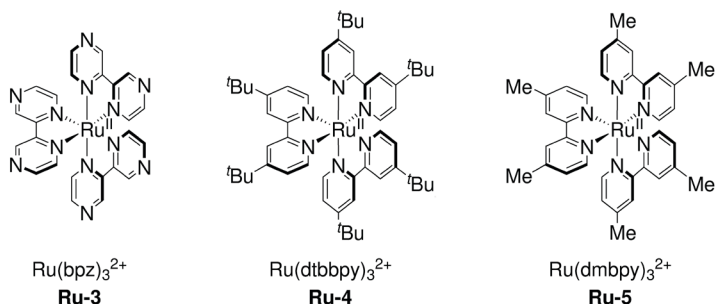
**Table 1.** Representative scope of enantioselective  $\alpha$ -alkylation of aldehydes using **Ru-1**

$\text{Ru}(\text{phen})_3^{2+}$  (**Ru-2**) as photocatalyst (Table 2).<sup>14</sup> Here, the more reducing excited state  $^*\text{Ru}(\text{phen})_3^{2+}$  can undergo SET with triflyl chloride, resulting in  $\bullet\text{CF}_3$  addition to an aromatic ring. This radical addition pathway results in an incredibly broad scope of successful aromatic substrates, including numerous pharmaceutical compounds such as Lipitor (not shown).



**Table 2.** Direct trifluoromethylation of arenes with  $\text{CF}_3\text{SO}_2\text{Cl}$  using **Ru-2**

Other analogues of  $\text{Ru}(\text{bpy})_3^{2+}$  have demonstrated broad applicability in organic synthesis, including those shown in Table 3. In particular,  $\text{Ru}(\text{bpz})_3^{2+}$  (**Ru-3**) has been used by the Yoon group to accomplish radical cation-mediated [4+2] cycloadditions of electronically-mismatched dienes and dienophiles,<sup>15</sup>

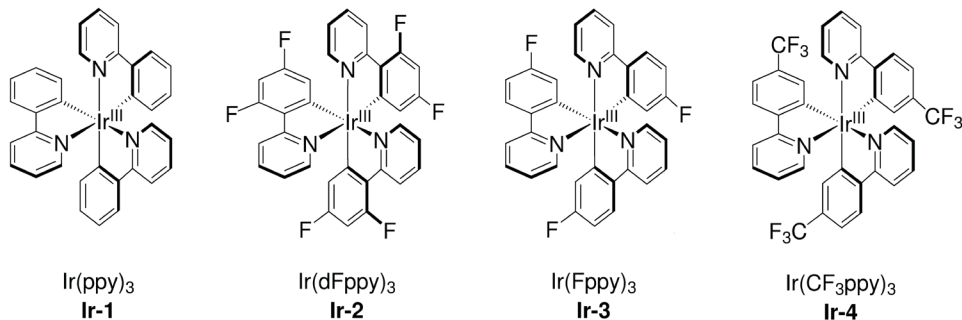


**Table 3.** Other ruthenium trisbipyridyl photocatalysts

while our group has used the same photocatalyst for the decarboxylative fluorination of certain alkyl carboxylic acids.<sup>16</sup> Ru(dtbbpy)<sub>3</sub><sup>2+</sup> (**Ru-4**) has also been used by Yoon and coworkers for the visible light sensitization of vinyl azides via energy transfer (ET) from the triplet excited state of the photocatalyst,<sup>17</sup> as well as by Rueping for the aerobic oxidation of benzylic alcohols to aldehydes and ketones.<sup>18</sup>

### Homoleptic Iridium Photocatalysts

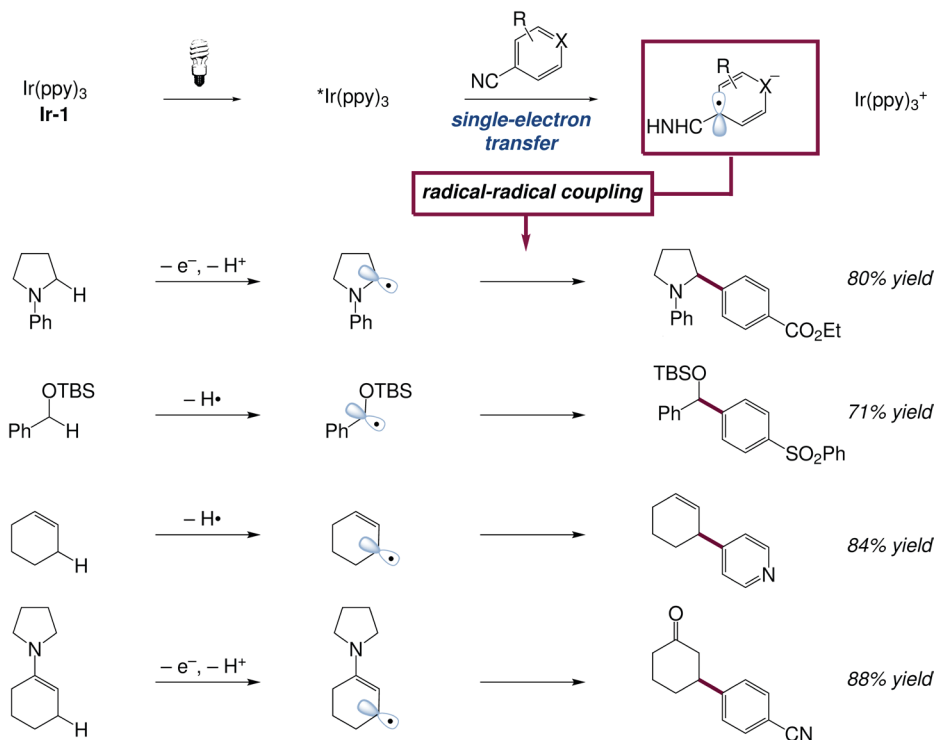
Owing to the ability to orthogonally manipulate the HOMO and LUMO energies of iridium polypyridyl complexes, a diverse suite of analogues of Ir(ppy)<sub>3</sub> have been developed for numerous uses in organic chemistry. In particular, homoleptic iridium photocatalysts, in which each ligand is the same cyclometalated phenylpyridine, have been utilized in transformations in which the excited photocatalyst performs a single-electron reduction of a substrate molecule, described as an oxidative quenching mechanism (Table 4).



**Table 4.** Homoleptic iridium photocatalysts

The parent molecule, Ir(ppy)<sub>3</sub> (**Ir-1**), previously used as a phosphorescent emitter in PhOLEDs, has been utilized extensively within our group's research program for its ability to accomplish challenging excited state reductions. In particular, the SET reduction of electron-deficient cyanoarenes, such as 1,4-dicyanobenzene, by \*Ir(ppy)<sub>3</sub> has enabled a number of radical-radical coupling reactions to generate arylated products via the intermediacy of persistent aryl radical anions (Scheme 2). In particular, our group has demonstrated the utility of this activation mode in the  $\alpha$ -arylation of amines via oxidation/deprotonation;<sup>19</sup>  $\alpha$ -arylation of benzylic ethers<sup>20</sup> and olefins<sup>21</sup> via thiyl radical-mediated hydrogen atom transfer; and  $\beta$ -arylation of carbonyls via enamine oxidation/deprotonation.<sup>22</sup> Our group also recently used **Ir-1** to enable the energy transfer-mediated esterification of aryl halides with carboxylic acids.<sup>23</sup>

Other homoleptic iridium photocatalysts, including those shown in Table 4, have been used in organic transformations by our group and others for their fine-tuned photophysical and electron-transfer properties. In particular, our group has used Ir(dFppy)<sub>3</sub> (**Ir-2**) as a highly competent complementary photocatalyst to Ru(phen)<sub>3</sub><sup>2+</sup> (**Ru-2**) in the arene C–H trifluoromethylation using triflyl chloride, while Alemán, Paton, and Smith have shown it to be an efficient photocatalyst for ET- induced radical cyclization reactions.<sup>24</sup> Meanwhile, the monofluorinated Ir(Fppy)<sub>3</sub> (**Ir-3**) has been shown to be an efficient photocatalyst for the asymmetric addition of  $\alpha$ -amino radicals into imines by Ooi,<sup>25</sup> while the trifluoromethyl analogue, Ir(CF<sub>3</sub>ppy)<sub>3</sub> (**Ir-4**) has been shown by Weaver to be efficient for defluorinative reactions of fluoroarenes.<sup>26</sup> These homoleptic iridium photocatalysts have received much attention for their ease of synthesis and broad applications.



**Scheme 2.** Radical-radical coupling arylation reactions using **Ir-1**

### Heteroleptic Iridium Photocatalysts

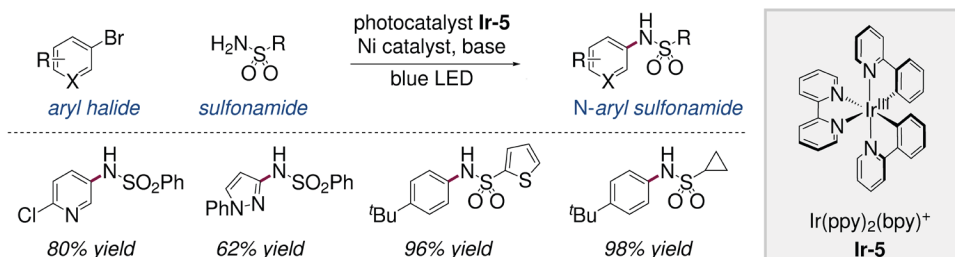
Cationic polypyridyl complexes of iridium(III), in which one of the phenylpyridine ligands is replaced by a bipyridine-type ligand, have been extensively used by synthetic organic chemists owing to the nearly complete orthogonality of the HOMO and LUMO, localized on the metal center and phenyl ring of the phenylpyridine, and bipyridine ligand, respectively. As such, the reducing and oxidizing power can be manipulated individually with minimal perturbation to the other.

#### Simple Heteroleptic Iridium Photocatalysts

Photocatalysts of the type  $\text{Ir(ppy)}_2(\text{N}^{\wedge}\text{N})^+$  have been exploited by our group and others for a variety of SET and ET-dependent transformations. The simplest heteroleptic iridium photocatalyst,  $\text{Ir(ppy)}_2(\text{bpy})^+$  (**Ir-5**) was recently used by our group in collaboration with Lee and coworkers as the ideal photocatalyst for an energy transfer-enabled metallaphotoredox sulfonamidation of aryl halides (Table 5).<sup>27</sup> In this case, the excited  $^*\text{Ir(III)}$  state of the photocatalyst could directly transfer its triplet energy to a Ni(II) aryl sulfonamido complex, leading to a highly efficient reductive elimination.

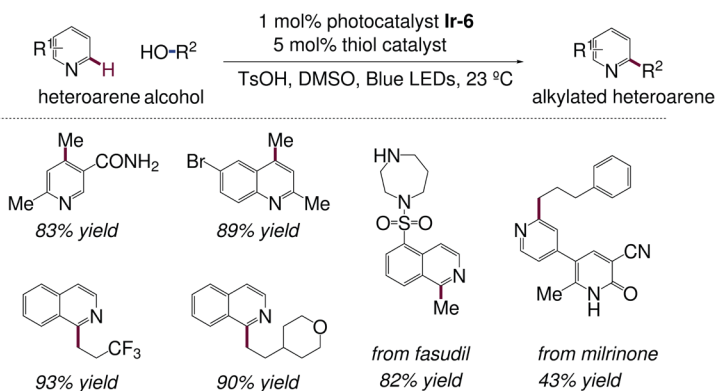
Furthermore, the di-*tert*-butyl-substituted analogue,  $\text{Ir(ppy)}_2(\text{dtbbpy})^+$  (**Ir-6**) has been utilized extensively by our group for a variety of transformations, including aldehyde  $\alpha$ -trifluoromethylation<sup>28</sup> and amine  $\alpha$ -heteroarylation.<sup>29</sup> One particularly interesting use of this photocatalyst is in the radical-radical coupling of  $\alpha$ -amino radicals formed by reduction of imines with other carbon-centered radicals formed through oxidation, such as enamine oxidation<sup>30</sup> and benzyl ether Hydrogen atom transfer (HAT).<sup>31</sup> In these cases, it is the fine-tuned oxidizing and reducing power of **Ir-6** that enables these transformations to work. Furthermore, **Ir-6** was found to be the ideal photocatalyst for the HAT-enabled spin-center shift-





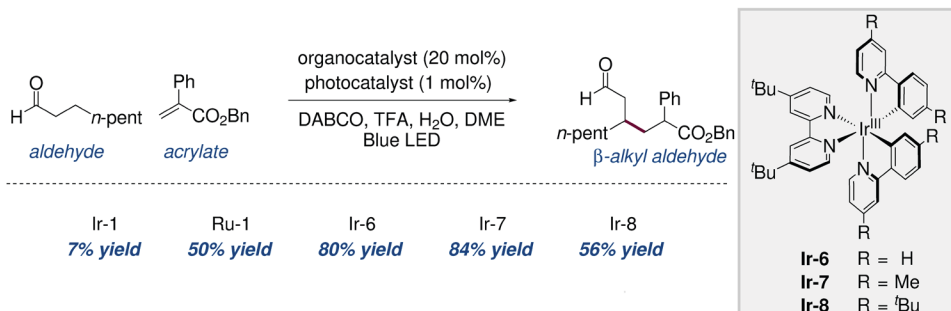
**Table 5.** Energy transfer-mediated sulfonamidation of aryl halides using Ir-5

mediated alkylation of heteroarenes with simple alcohols.<sup>32</sup> In this transformation, the oxidized Ir(IV) state of the photocatalyst can oxidize a thiol catalyst, which can subsequently abstract a hydrogen atom from an alcohol substrate. The resultant nucleophilic radical can add to a protonated heteroarene, which, after spin-center shift, generates an electron-deficient benzylic radical, which can be reduced by the excited state of the photocatalyst. This mechanism enables a variety of heteroarenes to be directly alkylated using simple alcohols, as shown in Table 6.



**Table 6.** Direct alkylation of heteroarenes with alcohols using Ir-6.

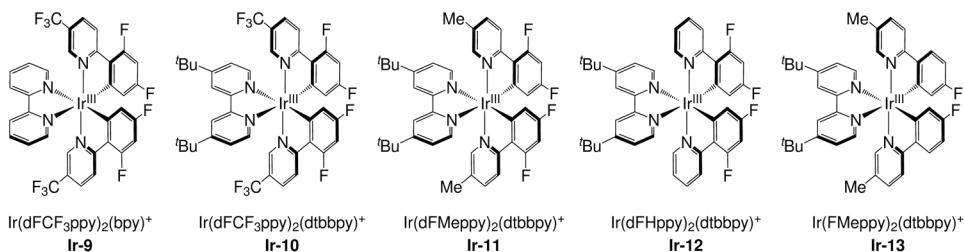
Further substitutions on the phenylpyridine ligand can be used to fine-tune reaction efficiency, as in the case the direct  $\beta$ -alkylation of aldehydes by addition of a catalytically-generated  $\beta$ -enaminyll radical to a Michael acceptor.<sup>33</sup> Indeed, as shown in Table 7, optimal yield of 80% could be obtained with Ir-6, while a diminished 56% yield was observed with Ir(dtbbpy)<sub>2</sub>(dtbbpy)<sup>+</sup> (Ir-8). However, a slight improvement in the yield to 84% yield could be obtained with Ir(dmppy)<sub>2</sub>(dtbbpy)<sup>+</sup> (Ir-7), leading to the optimized general conditions. This example demonstrates the effect that fine-tuning of photocatalyst structure and electronics can have on the efficiency of a desired transformation, necessitating a broad understanding of photocatalyst structure-function relationship for photoredox-mediated organic transformations.



**Table 7.** Direct  $\beta$ -alkylation of aldehydes via  $\beta$ -enaminyll radicals using **Ir-7**

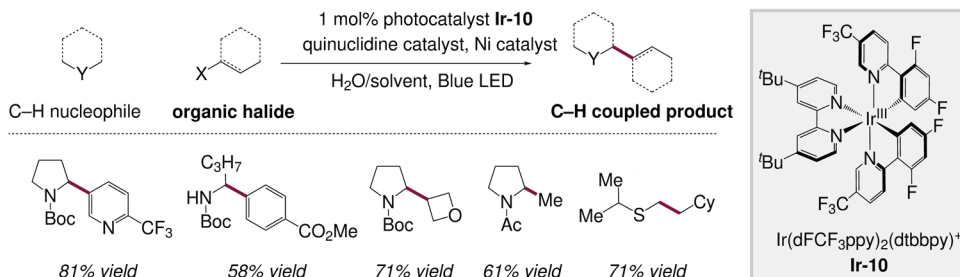
### Fluorinated Heteroleptic Iridium Photocatalysts

Owing to the orthogonal nature of the HOMO and LUMO of heteroleptic iridium photocatalysts, substitution of the phenyl ring of the phenylpyridine ligands can alter the HOMO energy level with minimal perturbation of the LUMO energy level, effectively shifting the oxidizing power without affecting the reducing power of the photocatalyst. Indeed, by substituting the phenylpyridine ligand with fluoro and trifluoromethyl groups, a number of more oxidizing photocatalysts can be prepared (Table 8). These more strongly oxidizing photocatalysts are capable of performing SET oxidations on functionalities such as carboxylates, amine, trifluoroborates, and silicates, among others.



**Table 8.** Iridium photocatalysts bearing fluorinated phenylpyridine ligands

In particular,  $\text{Ir}(\text{dFCF}_3\text{ppy})_2(\text{dtbbpy})^+$  (**Ir-10**) has been used extensively within our group in combination with nickel catalysis, enabling a number of transformations such as decarboxylative arylation,<sup>34</sup> alkylation,<sup>35</sup> and vinylation,<sup>36</sup> as well as alkylation,<sup>37</sup> etherification,<sup>38</sup> and amination<sup>39</sup> of aryl halides, while  $\text{Ir}(\text{dFCF}_3\text{ppy})_2(\text{bpy})^+$  (**Ir-9**) has been used by the Molander group for the SET-enabled transmetalation of trifluoroborates and silicates for similar cross-coupling reactions<sup>40</sup> and by the Knowles lab for alkene amidation via proton-coupled electron transfer.<sup>41</sup> As a representative example of the broad applicability of **Ir-10** in metallaphotoredox cross-couplings,<sup>42</sup> Table 9 displays the arylation<sup>43</sup> and alkylation<sup>44</sup> of hydric C–H bonds via the merger of HAT and metallaphotoredox catalysis. Here, the oxidizing nature of the excited state of the photocatalyst enables oxidation of the quinuclidine HAT catalyst,<sup>45</sup> while the reducing nature of the Ir(II) state allows for initial reduction of the Ni(II) precatalyst to the required Ni(0) oxidation state, as well as catalytic turnover by reduction of Ni(I) to Ni(0).



**Table 9.** C–H arylation and alkylation via HAT metallaphotoredox catalysis using **Ir-10**

In some cases, however, the highly electron-deficient phenylpyridine ligand of **Ir-9** and **Ir-10** proves detrimental to the overall efficiency of the reaction, oftentimes owing to direct addition of intermediate carbon-centered radicals to the electrophilic arenes. In these cases, catalysts Ir(dFMeppy)<sub>2</sub>(dtbbpy)<sup>+</sup> (**Ir-11**) and Ir(dFHppy)<sub>2</sub>(dtbbpy)<sup>+</sup> (**Ir-12**) can oftentimes be used to restore the efficiency of the reaction. For example, our group has shown that the decarboxylative vinylation of carboxylic acids with vinyl halides can be accomplished via metallaphotoredox catalysis.<sup>46</sup> As shown in Table 10, however, the use of photocatalyst **Ir-10** required dilute conditions, with insoluble inorganic base and high nickel catalyst loadings for optimal efficiency. If the reaction was run under more concentrated conditions with soluble organic base and lower nickel loadings, however, the maximum efficiency achieved was 61%. Under these conditions, substantial alkylated photocatalyst could be observed in the crude reaction mixture, stemming from direct radical addition to electrophilic sites on the phenyl pyridine ligand. Simply by exchanging the trifluoromethyl group for a methyl group (i.e. using **Ir-11** in place of **Ir-10**) led to a dramatic increase to the fully optimized 92% yield, demonstrating the value of the less electron-deficient dFMeppy ligand scaffold. A similar dramatic improvement in yield was observed between **Ir-10** and **Ir-11** in our group's direct aldehyde C–H alkylation transformation<sup>47</sup> and Knowles's intermolecular anti-Markovnikov hydroamination,<sup>48</sup> while **Ir-12** proved to be the ideal photocatalyst in our double-decarboxylative metallaphotoredox coupling of alcohol-derived oxalate esters.<sup>49</sup>

**α-oxy acid**      **vinyl iodide**      **allylic ether**

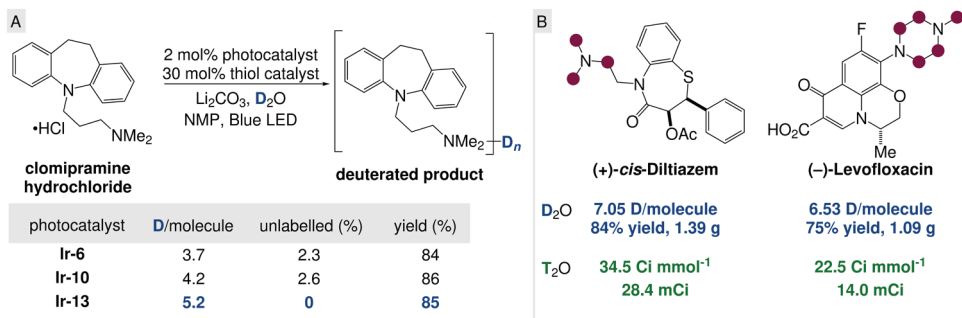
1 mol% photocatalyst, Ni catalyst  
 base, solvent, Blue LED

Ni loading	photocatalyst	base	solvent	time	yield
10 mol%	<b>Ir-10</b>	Cs <sub>2</sub> CO <sub>3</sub>	DMF (0.025 M)	72 h	83%
2 mol%	<b>Ir-10</b>	Cs <sub>2</sub> CO <sub>3</sub>	DMF (0.1 M)	18 h	22%
2 mol%	<b>Ir-10</b>	Cs <sub>2</sub> CO <sub>3</sub>	DMSO (0.1 M)	18 h	52%
2 mol%	<b>Ir-10</b>	DBU	DMSO (0.1 M)	18 h	61%
2 mol%	<b>Ir-11</b>	DBU	DMSO (0.1 M)	18 h	92%

**Table 10.** Superior decarboxylative vinylation of carboxylic acids using **Ir-11** vs. **Ir-10**

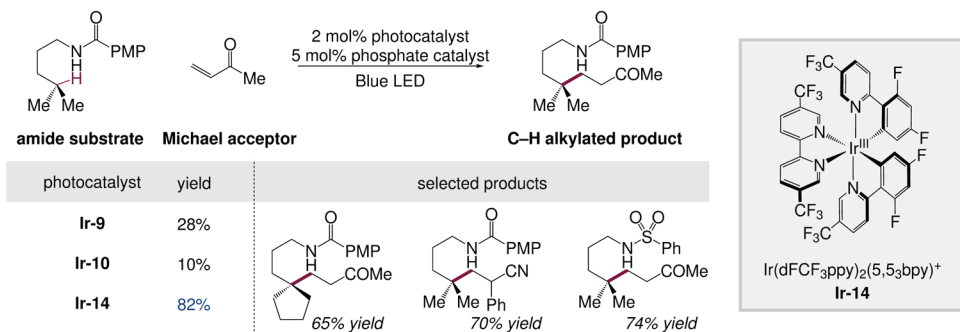
Furthermore, the slightly less-oxidizing Ir(FMeppy)<sub>2</sub>(dtbbpy)<sup>+</sup> (**Ir-13**) has found application in our group in two distinct transformations. Indeed, **Ir-13** has been used for the enantioselective alkylation of aldehydes with simple olefins<sup>50</sup> and the direct isotopic labeling of pharmaceutical molecules by Hydrogen Isotope Exchange (HIE).<sup>51</sup> Indeed, as shown in Table 11A, **Ir-13** was vastly superior to **Ir-10**,

delivering 5.2 D/molecule with 0% unlabeled substrate, whereas **Ir-10** delivered only 4.2 D/molecule with 2.6% unlabeled material remaining. Indeed, as shown in part in Table 11B, a number of pharmaceutical molecules could be successfully deuterated and tritiated at positions adjacent to oxidizable amines via the intermediacy of  $\alpha$ -amino radicals, using **Ir-13** as the photocatalyst.



**Table 11.** Photoredox HIE deuteration and tritiation of pharmaceutical compounds using **Ir-13**

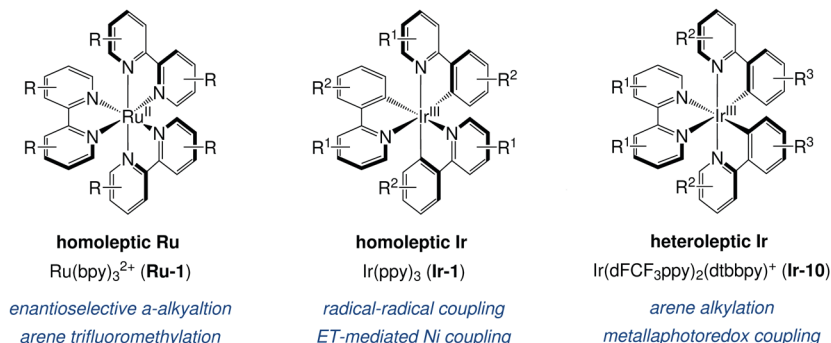
In addition to derivatization of the phenylpyridine substituents to modulate the oxidizing power of the photocatalyst, as in the **Ir-10–Ir-13** series, modifications to the bipyridine backbone, as between **Ir-9** and **Ir-10**, can be extended even further, to  $\text{Ir}(\text{dFCF}_3\text{ppy})_2(5,5'\text{-dCF}_3\text{bpy})$  (**Ir-14**). Here, **Ir-14** has severely diminished reductive capability, as the reduced Ir(II) state reduction potential is  $E_{1/2}^{\text{red}}(\text{Ir}^{\text{III}}/\text{Ir}^{\text{II}}) = -0.67 \text{ V}$  while that of **Ir-10** is  $E_{1/2}^{\text{red}}(\text{Ir}^{\text{III}}/\text{Ir}^{\text{II}}) = -1.37 \text{ V}$ , both vs. SCE. Indeed, the difference in these photocatalysts enabled Knowles's catalytic alkylation of remote C–H bonds via proton-coupled electron transfer (PCET)-enabled amidyl radical HAT (*Table 12*).<sup>52</sup>



**Table 12.** PCET-enabled remote C–H alkylation via amidyl radical abstraction using **Ir-14**

## Conclusions

In conclusion, the use of polypyridyl complexes of ruthenium and iridium as photocatalysts in organic transformations is a highly enabling mode of activating organic substrates towards SET and ET processes. The ability to use precisely tuned photocatalysts for the appropriate electrochemical potential or triplet energy requirement allows for the implementation of the ideal optimized reaction



**Table 13.** Representative Ru and Ir polypyridyl photocatalyst classes

conditions. Photocatalysts of the type Ru(N<sup>^</sup>N)<sub>3</sub><sup>2+</sup>, Ir(C<sup>^</sup>N)<sub>3</sub>, and Ir(C<sup>^</sup>N)<sub>2</sub>(N<sup>^</sup>N)<sup>+</sup> each have optimal uses in synthetic organic photocatalysis, as demonstrated by our group and others (Table 13). Indeed, by selecting the appropriate photocatalyst for the desired transformation, or extrapolating from known trends, optimal conditions can be developed. We anticipate that the implementation of the various photocatalysts described herein, and future iterations of these scaffolds, will greatly improve the scope of synthetic organic photocatalyzed transformations.

## Acknowledgments

Financial support provided by the NIHGMS (RO1 GM103558).

## References

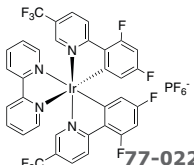
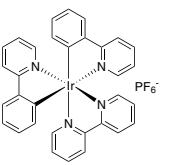
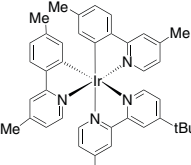
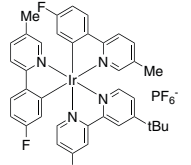
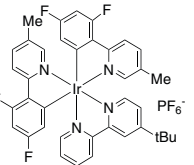
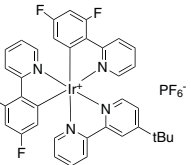
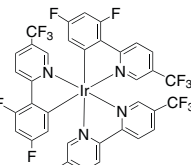
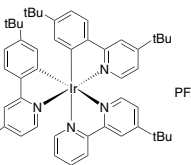
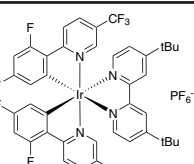
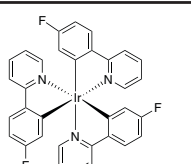
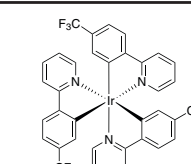
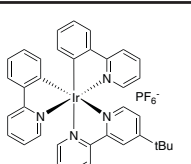
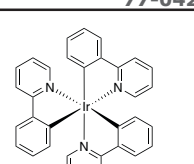
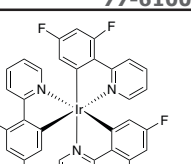
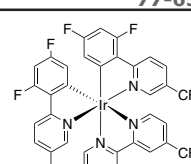
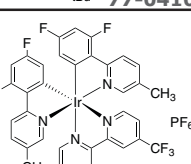
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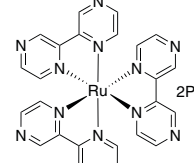
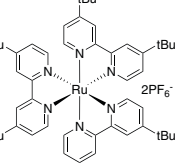
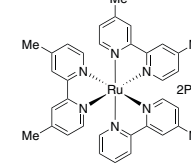
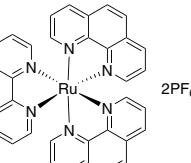
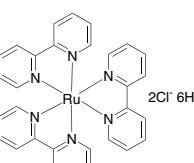
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# Photocatalysts - Quick Reference

## Iridium Photocatalysts

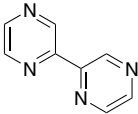
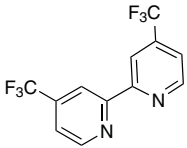
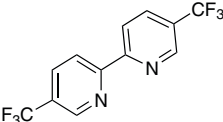
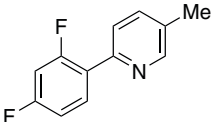
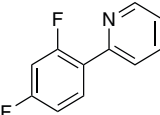
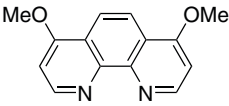
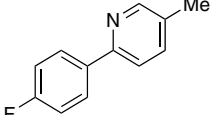
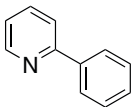
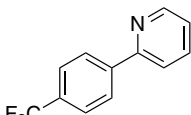
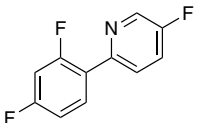
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## Ruthenium Photocatalysts

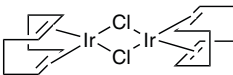
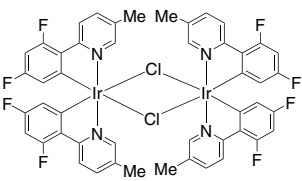
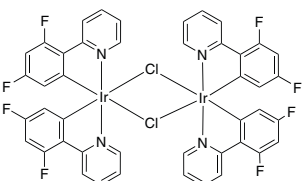
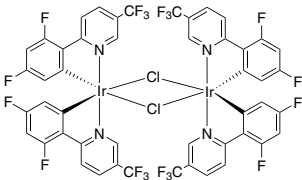
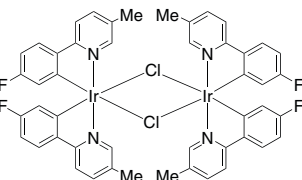
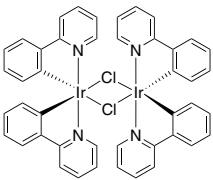
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# Photocatalysts - Quick Reference

## Ligands for Photocatalyst Synthesis

			
07-0750	07-1425	07-1430	07-1280
			
07-1420	07-1923	07-1410	07-1780
			
07-2625	07-1415		

## Precursors for Photocatalyst Synthesis

		
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77-0468	77-0335	77-0455



## IRIDIUM (Compounds)

77-0220

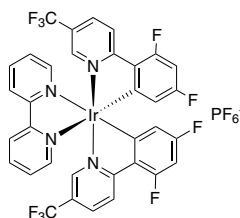
NEW

(2,2'-Bipyridine)bis[3,5-difluoro-2-[5-(trifluoromethyl)-2-pyridinyl-kN]][phenyl-kC] iridium(III) hexafluorophosphate, 95% (1092775-62-6)

C<sub>34</sub>H<sub>18</sub>F<sub>16</sub>IrN<sub>4</sub>P; FW: 1009.70; yellow powd.

air sensitive

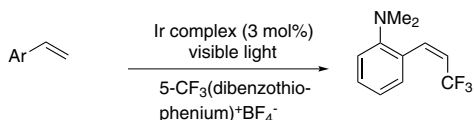
Note: Photocatalyst



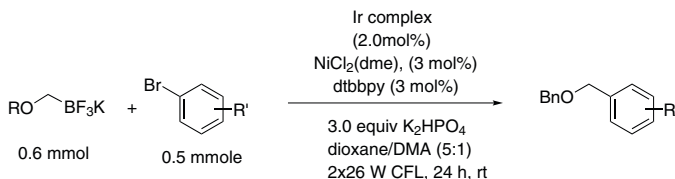
50mg  
250mg

### Technical Notes:

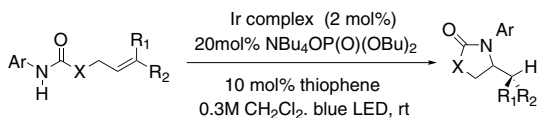
1. Photocatalyst used for the chemo-, regio-, and stereoselective trifluoromethylation of styrene.
2. Photoredox catalyst used in cross-coupling: Ir/Ni dual catalysts for the synthesis of benzylic ethers.
3. Iridium complex used for catalytic olefin hydroamidation enabled by proton-coupled electron transfer.
4. Catalyst used for visible light photoredox cross-coupling of acyl chlorides with potassium alkoxymethyltrifluoroborates.
5. Iridium catalyst used in the photoredox/nickel dual catalytic cross-coupling of secondary alkyl β-trifluoroborato ketones and esters with aryl bromides.
6. Photocatalyst used in the cross-coupling of trifluoroalkylboranes.



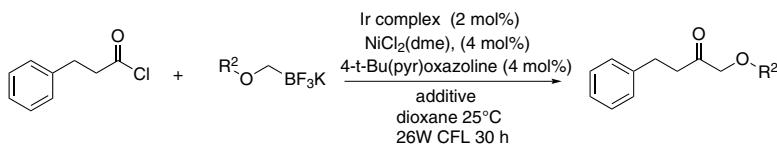
Tech. Note (1)  
Ref. (1)



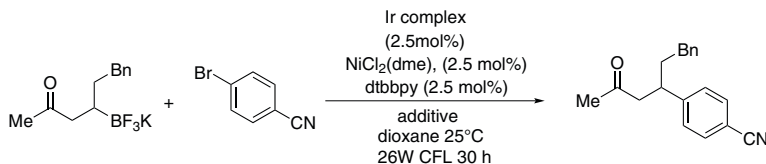
Tech. Note (2)  
Ref. (2)



Tech. Note (3)  
Ref. (3)



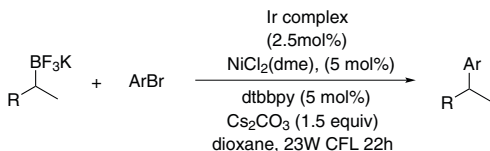
Tech. Note (4)  
Ref. (4)



Tech. Note (5)  
Ref. (5)

**IRIDIUM (Compounds)**

**77-0220** (2,2'-Bipyridine)bis[3,5-difluoro-2-[5-(trifluoromethyl)-2-pyridinyl-kN][phenyl-kC] iridium(III) hexafluorophosphate, 95% (1092775-62-6)

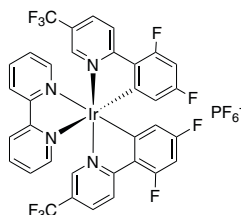


Tech. Note (6)  
Ref. (6)

References:

1. *J. Org. Chem.*, **2014**, *79*, 10446.
2. *Org. Lett.*, **2015**, *17*, 3294.
3. *J. Am. Chem. Soc.*, **2015**, *137*, 13495.
4. *Org. Lett.*, **2016**, *18*, 732.
5. *Org. Lett.*, **2016**, *18*, 2994.
6. *Org. Lett.*, **2016**, *18*, 5760.

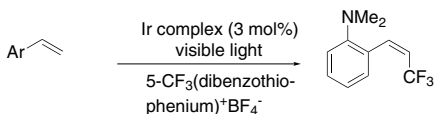
**77-0453** (2,2'-Bipyridine)bis[3,5-difluoro-2-[5-trifluoromethyl-2-pyridinyl-kN]phenyl-kC] iridium(III) hexafluorophosphate, 99% (1092775-62-6)  
[Ir(C<sub>10</sub>H<sub>8</sub>N<sub>2</sub>)(C<sub>12</sub>H<sub>5</sub>F<sub>5</sub>N)<sub>2</sub>] PF<sub>6</sub>; FW: 1009.70; yellow powdr.  
Note: Photocatalyst



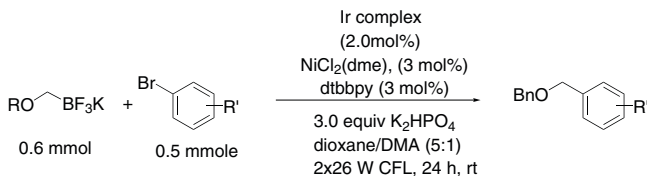
50mg  
250mg

Technical Notes:

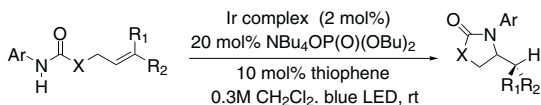
1. Catalyst used for the chemo-, regio, and stereoselective trifluoromethylation of styrene.
2. Photoredox catalyst used in cross-coupling: Ir/Ni dual catalysts for the synthesis of benzylic ethers.
3. Iridium complex used for catalytic olefin hydroamidation enabled by proton-coupled electron transfer.
4. Catalyst used for visible light photoredox cross-coupling of acyl chlorides with potassium alkoxymethyltrifluoroborates.
5. Iridium catalyst used in the photoredox/nickel dual catalytic cross-coupling of secondary alkyl β-trifluoroboratoketones and –esters with aryl bromides.
6. Photocatalyst used in the cross-coupling of trifluoroalkylboranes.



Tech. Note (1)  
Ref. (1)



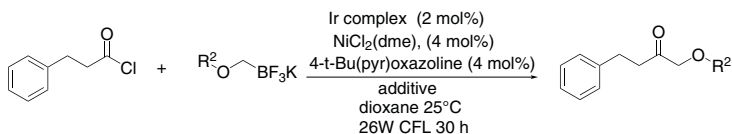
Tech. Note (2)  
Ref. (2)



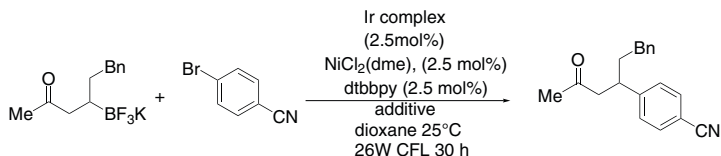
Tech. Note (3)  
Ref. (3)

**IRIDIUM (Compounds)**

**77-0453** (2,2'-Bipyridine)bis[3,5-difluoro-2-[5-(trifluoromethyl-2-pyridinyl-kN)phenyl-kC]iridium(III) hexafluorophosphate, 99% (1092775-62-6)



Tech. Note (4)  
Ref. (4)



Tech. Note (5)  
Ref. (5)



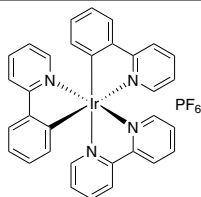
Tech. Note (6)  
Ref. (6)

References:

1. *J. Org. Chem.*, **2014**, *79*, 10446.
2. *Org. Lett.*, **2015**, *17*, 3294.
3. *J. Am. Chem. Soc.*, **2015**, *137*, 13495.
4. *Org. Lett.*, **2016**, *18*, 732.
5. *Org. Lett.*, **2016**, *18*, 2994.
6. *Org. Lett.*, **2016**, *18*, 5760.

**77-0465** (2,2'-Bipyridine)bis[2-pyridinyl-kN)phenyl-kC] iridium(III) hexafluorophosphate, 99% (106294-60-4)  
[Ir(C<sub>10</sub>H<sub>8</sub>N<sub>2</sub>)(C<sub>11</sub>H<sub>8</sub>N<sub>2</sub>)<sub>2</sub>]PF<sub>6</sub>; FW: 801.74; yellow powdr.  
Note: Photocatalyst

100mg  
500mg



Technical Notes:

1. Catalyst used in the visible-light, photoredox-catalyzed synthesis of nitrones.
2. Catalyst used in light-mediated, direct arylation of arenes and heteroarenes.
3. Photoredox catalyst used in C-P bond formation reactions.

References:

1. *Org. Lett.*, **2014**, *16*, 2872.
2. *Chem. Lett.*, **2013**, *42*, 1203.
3. *Chem. Comm.*, **2011**, *47*, 8679.

**IRIDIUM (Compounds)**

77-0218

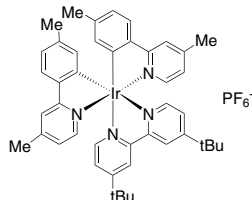
**NEW**

**4,4'-Bis(t-butyl-2,2'-bipyridine)bis[5-methyl-2-(4-methyl-2-pyridinyl-κN)phenyl-κC]iridium hexafluorophosphate, 95% (1607469-49-7)**

$C_{44}H_{48}F_6IrN_4P$ ; FW: 970.06; yellow pwr.

*air sensitive*

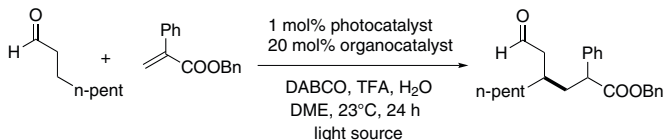
Note: Photocatalyst



50mg  
250mg

Technical Note:

- Catalyst used for the direct  $\beta$ -alkylation of aldehydes via photoredox organocatalysis.



**Tech. Note (1)  
Ref. (1)**

References:

- J. Am. Chem. Soc.*, **2014**, *136*, 6858.

77-0320

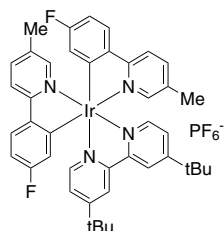
**NEW**

**[4,4'-Bis(1,1-dimethylethyl)-2,2'-bipyridine-κN,κN] bis[5-fluoro-2-(5-methyl-2-pyridinyl-κN)phenyl-κC] iridium hexafluorophosphate, 98% (808142-88-3)**

$C_{42}H_{42}F_6IrN_4P$ ; FW: 977.98; yellow solid

*air sensitive*

Note: Photocatalyst



50mg  
250mg

77-0330

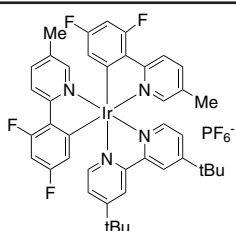
**NEW**

**[4,4'-Bis(1,1-dimethylethyl)-2,2'-bipyridine-κN,κN] bis[3,5-difluoro-2-(5-methyl-2-pyridinyl)phenyl] iridium hexafluorophosphate, 98% (1335047-34-1)**

$C_{42}H_{40}F_{10}IrN_4P$ ; FW: 1013.96; yellow solid

*air sensitive*

Note: Photocatalyst



100mg  
500mg

77-0350

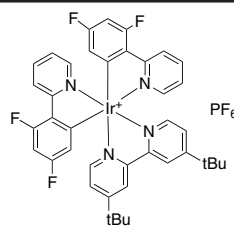
**NEW**

**[4,4'-Bis(1,1-dimethylethyl)-2,2'-bipyridine-κN,κN] bis[3,5-difluoro-2-(2-pyridinyl-κN)phenyl-κC]iridium hexafluorophosphate, 97% (1072067-44-7)**

$C_{40}H_{36}F_{10}IrN_4P$ ; FW: 985.92; Yellow pwr.

*air sensitive*

Note: Photocatalyst



100mg  
500mg

77-0380

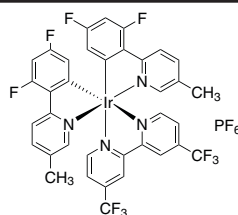
**NEW**

**4,4'-Bis(trifluoromethyl)-2,2'-bipyridinebis[3,5-difluoro-2-[5-methyl-2-pyridinyl)phenyl] iridium(III) hexafluorophosphate**

$C_{36}H_{22}F_{16}IrN_4P$ ; FW: 1037.77

*air sensitive*

Note: Photocatalyst



50mg  
250mg

**IRIDIUM (Compounds)**

77-0360

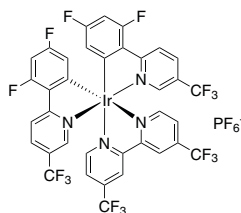
**NEW**

**4,4'-Bis(trifluoromethyl)-2,2'-bipyridinebis[3,5-difluoro-2-[5-trifluoromethyl-2-pyridinyl]phenyl]iridium(III) hexafluorophosphate (2030437-90-0)**

C<sub>36</sub>H<sub>16</sub>F<sub>22</sub>IrN<sub>4</sub>P; FW: 1145.69

*air sensitive*

Note: Photocatalyst



50mg  
250mg

77-0370

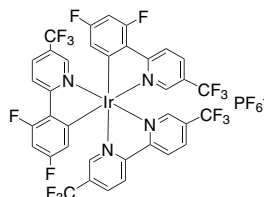
**NEW**

**[5,5'-Bis(trifluoromethyl)-2,2'-bipyridine-κN,κN]bis[3,5-difluoro-2-[5-(trifluoromethyl)-2-pyridinyl-κN]phenyl]iridium hexafluorophosphate, 98% (1973375-72-2)**

C<sub>36</sub>H<sub>16</sub>F<sub>22</sub>IrN<sub>4</sub>P; FW: 1145.69; yellow solid

*air sensitive*

Note: Photocatalyst



50mg  
250mg

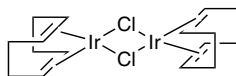
77-0400

**Chloro-1,5-cyclooctadiene iridium(I) dimer, 99% (12112-67-3)**

[IrCl(C<sub>8</sub>H<sub>12</sub>)<sub>2</sub>]<sub>2</sub>; FW: 671.71; red to orange powdr.;

m.p. 190° dec.

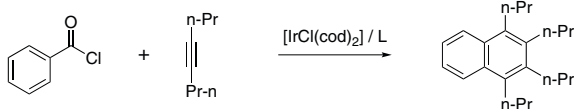
Note: Precursor for Photocatalyst Synthesis



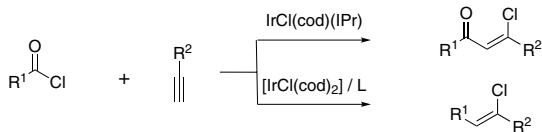
500mg  
2g  
10g

Technical Notes:

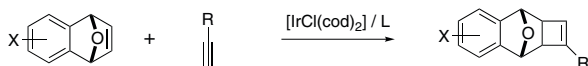
1. Precursor to catalysts for the asymmetric hydrogenation of tri- and tetrasubstituted olefins.
2. Precursor to catalyst for enantioselective reduction of imines.
3. Precursor to catalyst for allylic alkylation.
4. Precursor to catalyst for allylic amination and etherification.
5. Precursor to catalyst for the reaction of aryl chlorides with internal alkynes to produce substituted naphthalenes and anthracenes.
6. Ir-catalyzed addition of acid chlorides to terminal alkynes.
7. Intramolecular hydroamination of unactivated alkenes with secondary alkyl- and arylamines.
8. Enantioselective [2+2] cycloaddition.
9. Silyl-directed, Ir-catalyzed ortho-borylation of arenes.
10. Ir-catalyzed cross-coupling of styrene derivatives with allylic carbonates.
11. Transfer hydrogenative C-C coupling



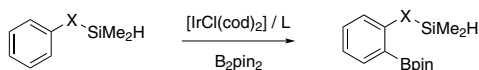
**Tech. Note (5)  
Ref. (5)**



**Tech. Note (6)  
Ref. (6)**



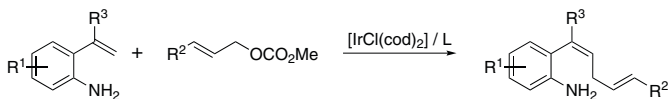
**Tech. Note (8)  
Ref. (8)**



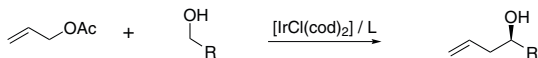
**Tech. Note (9)  
Ref. (9)**

**IRIDIUM (Compounds)**

**77-0400** Chloro-1,5-cyclooctadiene iridium(I) dimer, 99% (12112-67-3)  
(continued)



Tech. Note (10)  
Ref. (10)

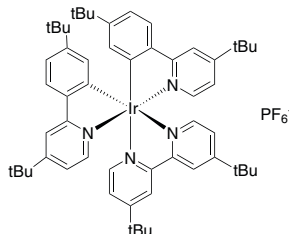


Tech. Note (11)  
Ref. (11)

References:

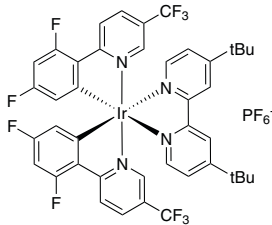
1. *Angew. Chem. Int. Ed.*, **1998**, 37, 2897
2. *J. Am. Chem. Soc.*, **1999**, 121, 6421
3. *J. Am. Chem. Soc.*, **1998**, 120, 8647
4. *J. Am. Chem. Soc.*, **2003**, 125, 14272
5. *J. Am. Chem. Soc.*, **2002**, 124, 12680
6. *J. Am. Chem. Soc.*, **2009**, 131, 6668
7. *J. Am. Chem. Soc.*, **2010**, 132, 413
8. *Org. Lett.*, **2010**, 12, 304
9. *J. Am. Chem. Soc.*, **2008**, 130, 7534
10. *J. Am. Chem. Soc.*, **2009**, 131, 8346
11. (a) *J. Am. Chem. Soc.*, **2008**, 130, 6340, (b) *Angew. Chem. Int. Ed.*, **2009**, 48, 6313

**77-0285** [4,4'-Di-t-butyl-2,2'-bipyridine][bis[5-(t-butyl)-2-[4-(t-butyl)-2-pyridinyl-kN]phenyl-kC]iridium(III) hexafluorophosphate, 95% (808142-80-5)  
C<sub>58</sub>H<sub>72</sub>F<sub>9</sub>IrN<sub>4</sub>P; FW: 1138.38; yellow powdr.  
air sensitive  
Note: Photocatalyst



50mg  
250mg

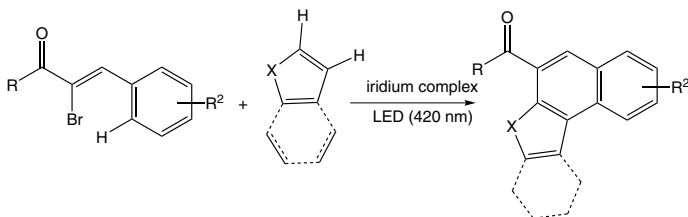
**77-0425** (4,4'-Di-t-butyl-2,2'-bipyridine)bis[3,5-difluoro-2-[5-trifluoromethyl-2-pyridinyl-kN]phenyl-kC]iridium(III) hexafluorophosphate, 99% (870987-63-6)  
[Ir(C<sub>18</sub>H<sub>24</sub>N<sub>2</sub>)(C<sub>12</sub>H<sub>8</sub>F<sub>3</sub>N<sub>2</sub>)<sub>2</sub>]<sup>+</sup>PF<sub>6</sub><sup>-</sup>; FW: 1121.91;  
yellow xtl.  
Note: Photocatalyst



50mg  
250mg  
1g

Technical Notes:

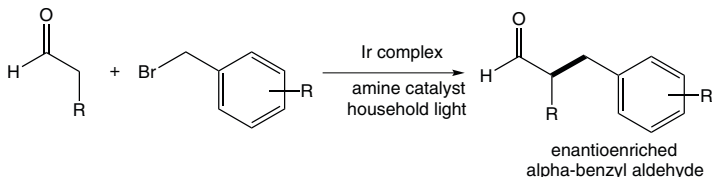
1. Visible light photoredox-catalyzed cascade cyclizations of  $\alpha$ -bromoaldehydes or  $\alpha$ -bromocinnamates with heteroarenes.
2. Enantioselective  $\alpha$ -benzylation of aldehydes via photoredox organocatalysis.



Tech. Note (1)  
Ref. (1)

**IRIDIUM (Compounds)**

**77-0425** (4,4'-Di-*t*-butyl-2,2'-bipyridine)bis[3,5-difluoro-2-[5-trifluoromethyl-2-pyridinyl- $\kappa$ N)phenyl- $\kappa$ C]iridium(III) hexafluorophosphate, 99% (870987-63-6)

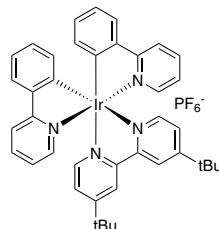


Tech. Note (2)  
Ref. (2)

References:

1. *Adv. Synth. Cat.*, **2014**, 356, 557
2. *J. Amer. Chem. Soc.*, **2010**, 132, 13600

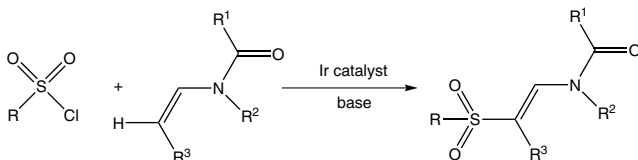
**77-0410** (4,4'-Di-*t*-butyl-2,2'-bipyridine)bis[2-(2-pyridinyl- $\kappa$ N)phenyl- $\kappa$ C]iridium(III) hexafluorophosphate, 99% (676525-77-2)  
[Ir(C<sub>18</sub>H<sub>24</sub>N<sub>2</sub>(C<sub>11</sub>H<sub>8</sub>N)<sub>2</sub>)<sub>2</sub>]<sup>+</sup>PF<sub>6</sub><sup>-</sup>; FW: 913.95; yellow xtl.  
Note: Photocatalyst



100mg  
500mg

Technical Notes:

1. This Iridium catalyst is used in the synthesis of  $\beta$ -amidovinyl sulfones via visible-light photoredox catalysis.
2. Numerous uses of this photoredox catalyst are reported (see Ref. 2).

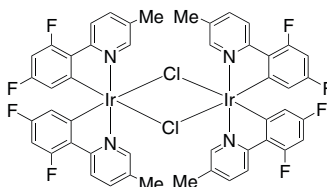


Tech. Note (1)  
Ref. (1)

References:

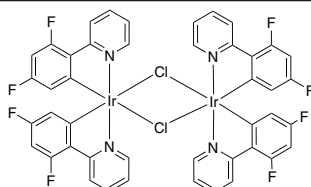
1. *Adv. Synth. Cat.*, **2013**, 355, 809
2. *Chem. Rev.*, **2013**, 113, 5322, review

**77-0345** Di- $\mu$ -chlorotetrakis[3,5-difluoro-2-(5-methyl-2-pyridinyl- $\kappa$ N)phenyl- $\kappa$ C]diiridium, 98% (1335047-33-0)  
C<sub>48</sub>H<sub>32</sub>Cl<sub>2</sub>F<sub>8</sub>Ir<sub>2</sub>N<sub>4</sub>; FW: 1272.11; yellow solid  
Note: Precursor for Photocatalyst Synthesis



250mg  
1g

**77-0365** Di- $\mu$ -chlorotetrakis[3,5-difluoro-2-(2-pyridinyl- $\kappa$ N)phenyl- $\kappa$ C]diiridium, 98% (562824-27-5)  
C<sub>44</sub>H<sub>24</sub>Cl<sub>2</sub>F<sub>8</sub>Ir<sub>2</sub>N<sub>4</sub>; FW: 1216.05; yellow solid  
*air sensitive*  
Note: Precursor for Photocatalyst Synthesis



250mg  
1g

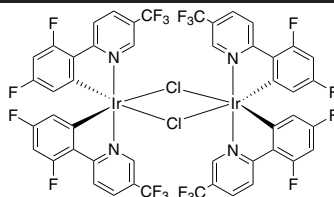
**IRIDIUM (Compounds)**

77-0468

NEW

**Di- $\mu$ -chlorotetrakis[3,5-difluoro-2-[5-trifluoromethyl-2-pyridinyl- $\kappa$ N)]phenyl- $\kappa$ C]diiridium(III), 99%** (870987-64-7)

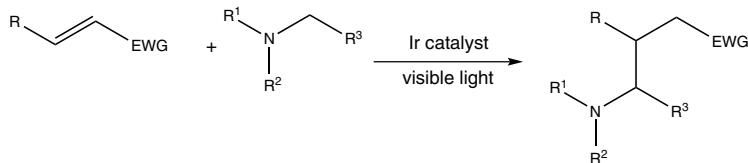
$C_{48}H_{20}Cl_2F_{20}Ir_2N_4$ ; FW: 1488.01; yellow xtl.  
Note: Precursor for Photocatalyst Synthesis



50mg  
250mg

Technical Note:

1. Addition to electron-deficient alkenes using a photoredox catalyst.



Tech. Note (1)  
Ref. (1)

References:

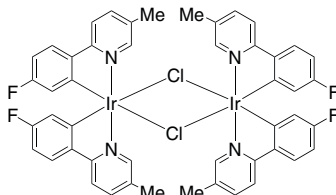
1. *J. Am. Chem. Soc.*, **2012**, *134*, 3338.

77-0335

NEW

**Di- $\mu$ -chlorotetrakis[5-fluoro-2-(5-methyl-2-pyridinyl- $\kappa$ N)]phenyl- $\kappa$ C]diiridium, 98%** (808142-89-4)

$C_{48}H_{36}Cl_2F_4Ir_2N_4$ ; FW: 1200.15; yellow solid  
Note: Precursor for Photocatalyst Synthesis

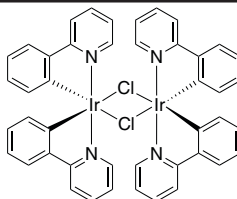


250mg  
1g

77-0455

**Di- $\mu$ -chlorotetrakis[2-(2-pyridinyl- $\kappa$ N)]phenyl- $\kappa$ C]diiridium(III), 99%** (603109-48-4)

$C_{44}H_{32}Cl_2Ir_2N_4$ ; FW: 1072.09; yellow-green xtl.  
Note: Precursor for Photocatalyst Synthesis



250mg  
1g

Technical Note:

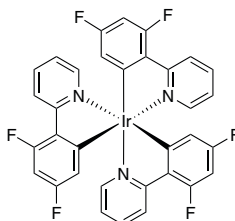
1. Iridium complex is a photoredox catalyst having numerous uses in electroluminescent materials and devices, organic light-emitting diodes, display devices and chemosensors.

77-7030

NEW

**Tris[2-(2,4-difluorophenyl)pyridine]iridium(III), 95%** (387859-70-3)

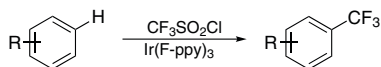
$C_{33}H_{18}F_6IrN_3$ ; FW: 762.72; yellow pwdr.  
*air sensitive*  
Note: Photocatalyst



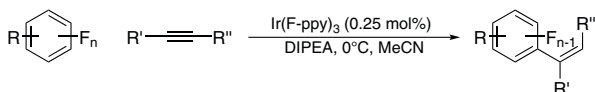
50mg  
250mg

Technical Notes:

1. Photoredox catalysis for trifluoromethylation of arenes and heteroarenes.
2. Photocatalyst for C-F alkenylation coupling reactions between perfluoroarenes and alkynes.



Tech. Note (1)  
Ref. (1)



Tech. Note (2)  
Ref. (2)

References:

1. *Nature*, **2011**, *480*, 224.
2. *Chem. Sci.*, **2016**, *7*, 6796.



**IRIDIUM (Compounds)**

77-6100

**NEW**

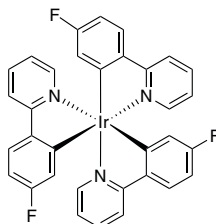
**Tris[5-fluoro-2-(2-pyridinyl-kN)phenyl-kC] iridium(III), 95%** (370878-69-6)

C<sub>33</sub>H<sub>21</sub>F<sub>3</sub>IrN<sub>3</sub>; FW: 708.75; yellow powdr.

air sensitive

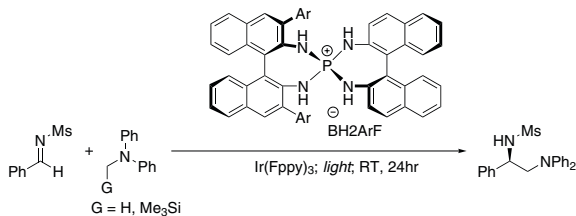
Note: Photocatalyst

50mg  
250mg

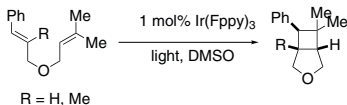


Technical Notes:

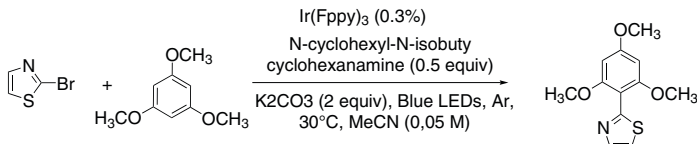
1. Photosensitizer for the enantioselective coupling reaction aldimines and (N-methanesulfonyl)-aldimines catalyzed by P-Spiro chiral (arylamino)phosphonium catalyst.
2. Photocatalyst for [2+2] styrene cycloadditions.
3. Photocatalyst for aryloxylation of trimethoxybenzene by via C-H functionalization.



**Tech. Note (1)**  
**Ref. (1)**



**Tech. Note (2)**  
**Ref. (2)**



**Tech. Note (3)**  
**Ref. (3)**

References:

1. *J. Org. Chem.*, **2016**, *81*, 6953.
2. *Chem. Sci.*, **2016**, *7*, 6796.
3. *Org. Lett.*, **2016**, *18*, 3996.

77-7015

**NEW**

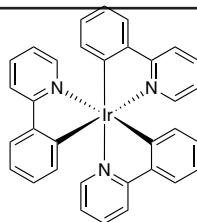
**Tris(2-phenylpyridinato-C2,N)iridium(III), 95%** (94928-86-8)

C<sub>33</sub>H<sub>24</sub>IrN<sub>3</sub>; yellow powdr.

air sensitive

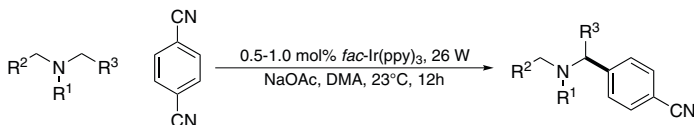
Note: Photocatalyst

50mg  
250mg



Technical Notes:

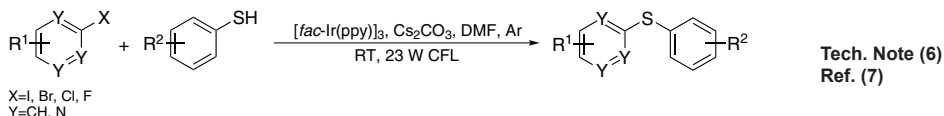
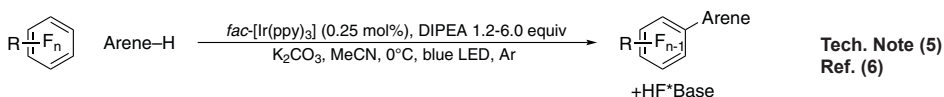
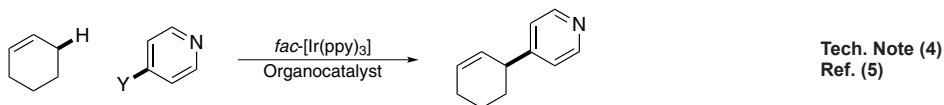
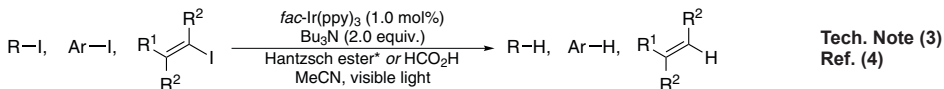
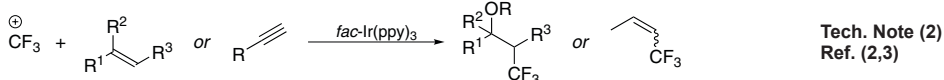
1. Photocatalyst for α-amino C-H arylation of cyano(hetero)arenes by tertiary amines
2. Photocatalyst for trifluoromethylation of alkenes and alkynes
3. Photocatalyst for reduction of alkyl, alkenyl, aryl iodides (a) and intramolecular reductive cyclizations (d)
4. Photocatalyst for organocatalyst assisted direct arylation of allylic sp<sup>3</sup> C-H bonds
5. Photocatalyst for the generation multifluorinated biaryls via functionalization of the C-F bond of a perfluoroarene and C-H bond of the other arene in the presence of amines
6. Photocatalyst for visible-light photoredox arylation of thiols with various aryl halides



**Tech. Note (1)**  
**Ref. (1)**

**IRIDIUM (Compounds)**

**77-7015** Tris(2-phenylpyridinato-C2,N)iridium(III), 95% (94928-86-8)  
(continued)

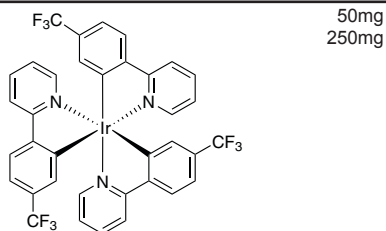


References:

1. *Science* **2011**, 334, 1114
2. *Angew. Chem. Int. Ed.* **2012**, 51, 9567
3. *Angew. Chem. Int. Ed.* **2014**, 53, 539
4. *Nat. Chem.* **2012**, 4, 854
5. *Nature* **2015** 519, 74
6. *J. Am. Chem. Soc.* **2016**, 138, 2520
7. *Angew. Chem. Int. Ed.* **2017**, 56, 874

**77-6580** Tris[(2-(2-pyridinyl-kN)-5-(trifluoromethyl) phenyl-kC]iridium(III), 95% (500295-52-3)  
 $\text{C}_{36}\text{H}_{21}\text{F}_9\text{IrN}_3$ ; FW: 858.78; yellow solid  
air sensitive  
Note: Photocatalyst

NEW



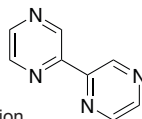
**NITROGEN (Compounds)**

07-0750

**2,2'-Bipyrazine, 95% (10199-00-5)**

C<sub>8</sub>H<sub>6</sub>N<sub>4</sub>; FW: 158.16; light-brown solid  
air sensitive

Note: Ligand for Photocatalyst Synthesis



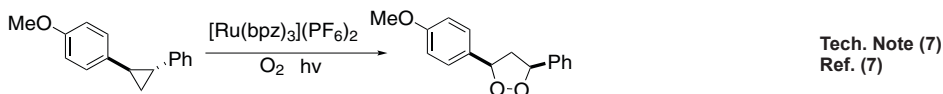
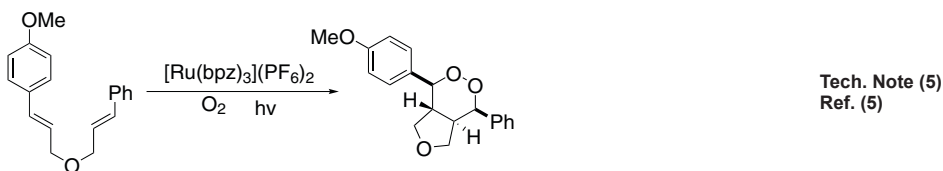
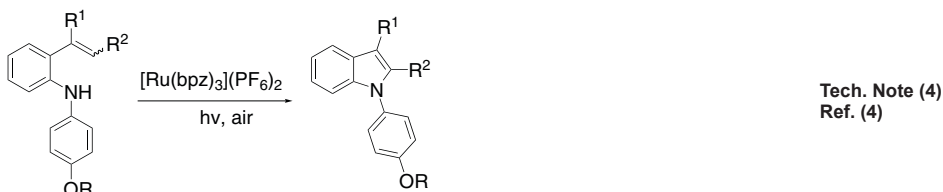
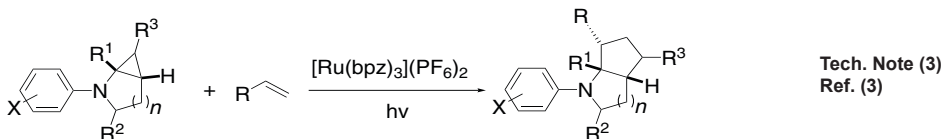
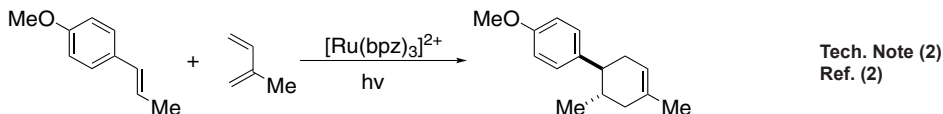
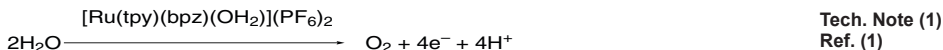
250mg

1g

**NEW**

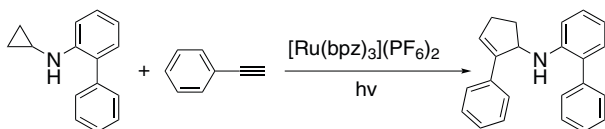
Technical Notes:

1. Ligand for the ruthenium- promoted catalytic water oxidation reaction.
2. Ligand for the ruthenium promoted photocatalytic Diels-Alder cycloaddition.
3. Ligand for the ruthenium photocatalyzed intermolecular [3+2] cycloaddition of cyclopropylamines with olefins.
4. Ligand for the ruthenium mediated photocatalytic reaction for the preparation of N-arylimidoles.
5. Endoperoxide synthesis by photocatalytic aerobic [2+2+2] cycloadditions.
6. [Ru(bpz)<sub>3</sub>](PF<sub>6</sub>)<sub>2</sub> catalyzed anti-Markovnikov hydrothiolation of olefins with a variety of thiols.
7. [Ru(bpz)<sub>3</sub>](PF<sub>6</sub>)<sub>2</sub> catalyzed [3+2] photooxygenation of aryl cyclopropanes.
8. [Ru(bpz)<sub>3</sub>](PF<sub>6</sub>)<sub>2</sub> catalyzed intermolecular [3 + 2] annulation of cyclopropylanilines with alkynes.



**NITROGEN (Compounds)**

**07-0750** 2,2'-Bipyrazine, 95% (10199-00-5)  
(continued)



Tech. Note (8)  
Ref. (8)

References:

1. *J. Am. Chem. Soc.*, **2008**, *130*, 16462.
2. *J. Am. Chem. Soc.*, **2011**, *133*, 19350.
3. *Angew. Chem. Int. Ed.*, **2012**, *51*, 222.
4. *Angew. Chem. Int. Ed.*, **2012**, *51*, 9562.
5. *Org. Lett.*, **2012**, *14*, 1640.
6. *J. Org. Chem.*, **2013**, *78*, 2046.
7. *Tetrahedron*, **2014**, *70*, 4270.
8. *Beilstein J. Org. Chem.*, **2014**, *10*, 975.

**07-1425** 4,4'-Bis(trifluoromethyl)-2,2'-bipyridine, min. 95%

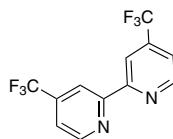
**NEW**

(142946-79-0)

C<sub>12</sub>H<sub>6</sub>F<sub>6</sub>N<sub>2</sub>; FW: 292.17; off-white to light yellow powdr.

air sensitive

Note: Ligand for Photocatalyst Synthesis



1g  
5g

**07-1430** 5,5'-Bis(trifluoromethyl)-2,2'-bipyridine, min 97%

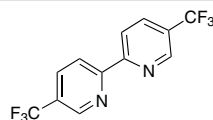
**NEW**

(142946-80-3)

C<sub>12</sub>H<sub>6</sub>F<sub>6</sub>N<sub>2</sub>; FW: 292.17; White powdr.

air sensitive

Note: Ligand for Photocatalyst Synthesis



1g  
5g

**07-1280** 2-(2,4-Difluorophenyl)-5-methylpyridine, 95%

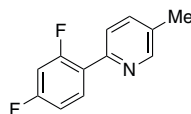
**NEW**

(583052-21-5)

C<sub>12</sub>H<sub>9</sub>F<sub>2</sub>N; FW: 205.20; white solid

air sensitive

Note: Ligand for Photocatalyst Synthesis



500mg  
2g

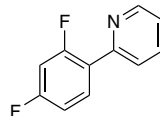
**07-1420** 2-(2,4-Difluorophenyl)pyridine, min. 97% (391604-55-0)

**NEW**

C<sub>11</sub>H<sub>7</sub>F<sub>2</sub>N; FW: 191.17; white solid

air sensitive

Note: Ligand for Photocatalyst Synthesis



1g  
5g

**07-1923** 4,7-Dimethoxy-1,10-phenanthroline, 98% (92149-07-0)

**NEW**

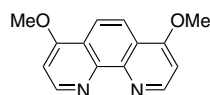
HAZ

C<sub>14</sub>H<sub>10</sub>N<sub>2</sub>O<sub>2</sub>; FW: 238.24; white to off-white powdr.;

m.p. 210-212°; d. 1.25

air sensitive

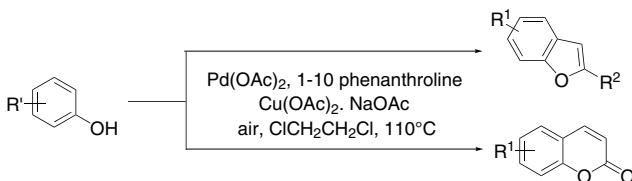
Note: Ligand for Photocatalyst Synthesis



250mg  
1g

Technical Notes:

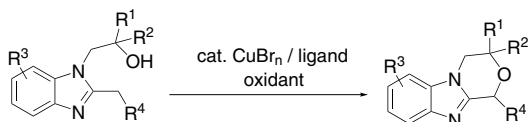
1. Palladium-catalyzed synthesis of benzofurans and coumarins from phenols and olefins.
2. Copper-catalyzed benzylic C(sp<sup>3</sup>)-H alkoxylation of heterocyclic compounds.
3. Synthesis of amides via copper-catalyzed amidation of aryl halides using isocyanides.
4. Iridium-catalyzed silylation of aryl C-H bonds.
5. Palladium-catalyzed intramolecular cyclization of nitroalkenes: synthesis of thienopyrroles.
6. A Copper-catalyzed N-alkynylation route to 2-substitued N-alkynyl pyrroles and their cyclization into pyrrolo[2,1-c]oxazin-1-ones



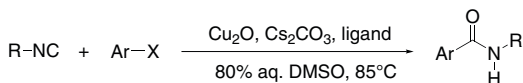
Tech. Note (1)  
Ref. (1)

**NITROGEN (Compounds)**

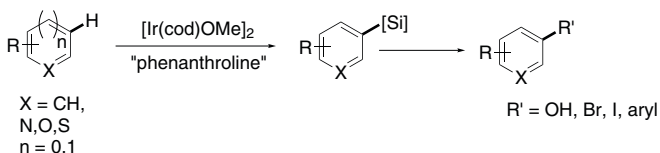
07-1923 4,7-Dimethoxy-1,10-phenanthroline, 98% (92149-07-0)  
(continued)



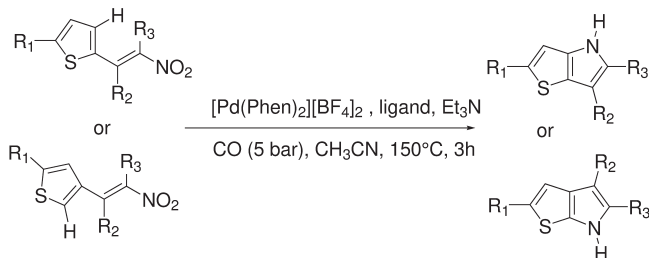
Tech. Note (2)  
Ref. (2)



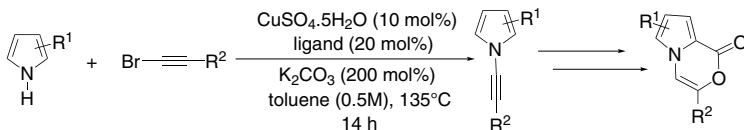
Tech. Note (3)  
Ref. (3)



Tech. Note (4)  
Ref. (4)



Tech. Note (5)  
Ref. (5)



Tech. Note (6)  
Ref. (6)

References:

1. *Angew. Chem. Int. Ed.*, **2013**, 52, 12669.
2. *Organic & Biomolecular Chemistry*, **2014**, 12, 2528.
3. *Tetrahedron Letts.*, **2014**, 55, 4981.
4. *J. Am. Chem. Soc.*, **2015**, 137, 592.
5. *European Journal of Organic Chemistry*, **2017**, 2017(14), 1902.
6. *Synthesis*, **2017**, 49, 2544.

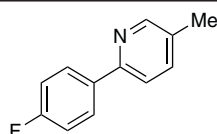
07-1410 2-(4-Fluorophenyl)-5-methylpyridine, min. 97%

NEW

(85237-65-6)  
C<sub>12</sub>H<sub>10</sub>FN; FW: 187.07; Off white powdr.

air sensitive

Note: Ligand for Photocatalyst Synthesis



1g  
5g

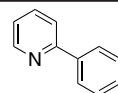
07-1780 2-Phenylpyridine, 95% (1008-89-5)

NEW

C<sub>11</sub>H<sub>9</sub>N; FW: 155.20; amber liquid; b.p. 268-270°; f.p. 230°;  
d. 1.086

air sensitive

Note: Ligand for Photocatalyst Synthesis



1g

## NITROGEN (Compounds)

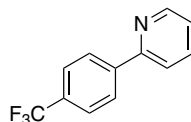
07-2625

2-[4-(Trifluoromethyl)phenyl]pyridine, 95%  
(203065-88-7)

1g

NEW

$C_{12}H_8F_3N$ ; FW: 223.19; white to yellow solid  
*air sensitive*  
Note: Ligand for Photocatalyst Synthesis



## RUTHENIUM (Compounds)

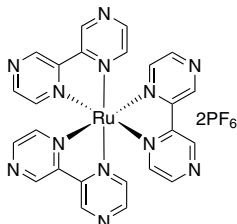
44-7910

Tris(2,2'-bipyrazine)ruthenium(II)  
hexafluorophosphate, 95% (80907-56-8)

50mg  
250mg

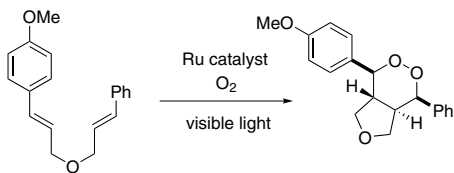
NEW

$C_{24}H_{18}F_{12}N_{12}P_2Ru$ ; FW: 865.48; red powder.  
*air sensitive*  
Note: Photocatalyst.

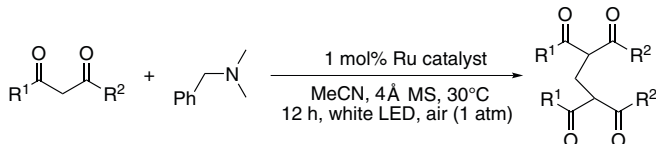


### Technical Notes:

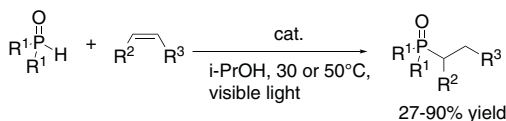
1. Endoperoxide synthesis by photocatalytic aerobic [2+2+2] cycloadditions.
2. Aerobic oxidation of a tertiary aliphatic amine under visible-light photocatalysis. Facile synthesis of methylene-bridged bis-1,3-dicarbonyl compounds.
3. Hydrophosphinylation of unactivated alkenes with secondary phosphine oxides under visible-light photocatalysis.
4. [3+2] Photooxygenation of aryl cyclopropanes via visible light photocatalysis.
5. Photocatalytic synthesis of dihydrobenzofurans by oxidative cycloaddition of phenols.



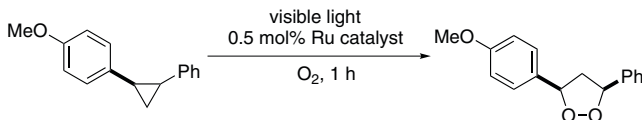
Tech. Note (1)  
Ref. (1)



Tech. Note (2)  
Ref. (2)



Tech. Note (3)  
Ref. (3)



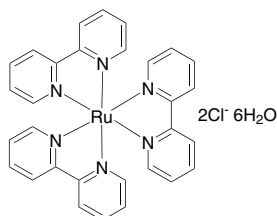
Tech. Note (4)  
Ref. (4)

### References:

1. *Org. Lett.*, **2012**, *14*, 1640.
2. *Chemistry – An Asian Journal*, **2012**, *7*, 2764.
3. *Green Chemistry*, **2013**, *15*, 1844.
4. *Advanced Synthesis & Catalysis*, **2014**, *356*, 2831.
5. *J. Am. Chem. Soc.*, **2015**, *137*, 5654.

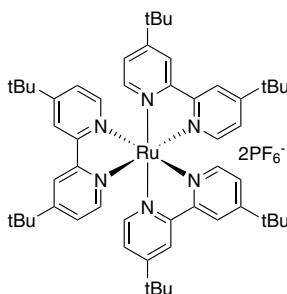
**RUTHENIUM (Compounds)**

**44-7900** **Tris(2,2'-bipyridyl)ruthenium(II) chloride hexahydrate, min. 98%** (50525-27-4)  
 $\text{Ru}(\text{C}_{10}\text{H}_8\text{N}_2)_3\text{Cl}_2 \cdot 6\text{H}_2\text{O}$ ; FW: 640.54 (748.63); orange to red xtl.  
 Note: Photocatalyst



250mg  
1g  
5g

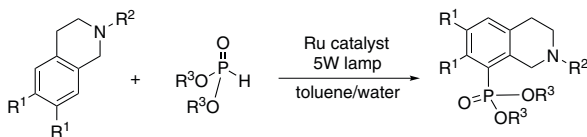
**44-7940** **Tris[4,4'-bis(t-butyl)-2,2'-bipyridine] ruthenium(II) hexafluorophosphate, 95%** (75777-87-6)  
 $\text{C}_{54}\text{H}_{72}\text{F}_{12}\text{N}_6\text{RuP}_2$ ; FW: 1196.19; red powdr.  
*air sensitive*  
 Note: Photocatalyst.



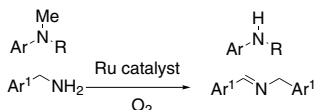
50mg  
250mg

Technical Notes:

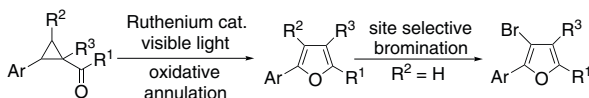
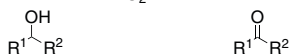
- Photoredox catalysed C-P bond formation reactions – visible light mediated oxidative phosphorylations of amines.
- Aerobic oxidation of a tertiary aliphatic amine under visible-light photocatalysis: facile synthesis of methylene-bridged bis-dicarbonyl compounds.
- Photoredox catalysis as an efficient tool for the aerobic oxidation of amines and alcohols.
- Visible-light induced, direct synthesis of polysubstituted furans from cyclopropyl ketones.



Tech. Note (1)  
Ref. (1)



Tech. Note (2)  
Ref. (2)



Tech. Note (3)  
Ref. (3)

References:

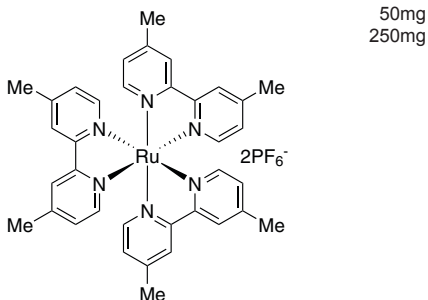
- Chem. Commun.*, **2011**, 47, 8679.
- Chemistry – An Asian Journal*, **2012**, 7, 2764.
- ACS Catalysis*, **2012**, 2, 2810.
- J. Org. Chem.*, **2016**, 81, 7008.

**RUTHENIUM (Compounds)**

44-7930

**NEW**

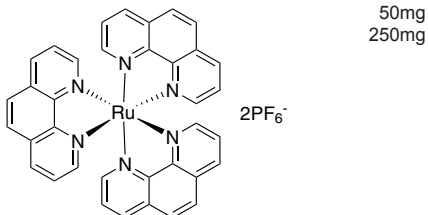
**Tris(4,4'-dimethyl-2,2'-bipyridine) ruthenium(II) hexafluorophosphate, 95%, DMBPY (83605-44-1)**  
 $C_{36}H_{36}F_{12}N_6RuP_2$ ; FW: 943.71; red pwdr.  
*air sensitive*  
 Note: Photocatalyst.



44-7955

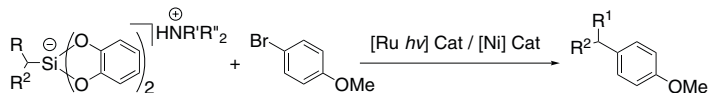
**NEW**

**Tris(1,10-phenanthroline)ruthenium(II) hexafluorophosphate, 95% (60804-75-3)**  
 $C_{36}H_{24}F_{12}N_6RuP_2$ ; FW: 931.62; red pwdr.  
*air sensitive*  
 Note: Photocatalyst

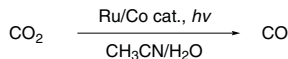


Technical Notes:

1. Photoredox catalyst for nickel assisted cross-coupling reactions of ammonium alkylsilicates with aryl bromides
2. A photosensitizer for cobalt catalyzed visible-light driven  $CO_2$  - Reduction to CO in  $CH_3CN/H_2O$  Solution



**Tech. Note (1)**  
**Ref. (1)**



**Tech. Note (2)**  
**Ref. (2)**

References:

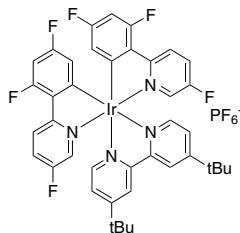
1. *J. Am. Chem. Soc.*, **2016**, 138, 475.
2. *Angew. Chem. Int. Ed.*, **2017**, 56, 738.

**COMING SOON...**

77-0340

**NEW**

**[4,4'-Bis(1,1-dimethylethyl)-2,2'-bipyridine]bis[3,5-difluoro-2-(5-fluoro-2-pyridinyl)phenyl]iridium hexafluorophosphate (2042201-18-1)**  
 $C_{40}H_{34}F_{12}IrN_4P$ ; FW: 1021.89  
*air sensitive*  
 Note: Photocatalyst





**KITS - Iridium Photocatalyst Kit 1**

96-7780

**Iridium Photocatalyst Kit 1**

Components also available for individual sale.

Contains the following:

**NEW**

	77-0218	50mg	
	77-0285	50mg	
	77-0410	100mg	
	77-0425	50mg	
	77-0453	50mg	
	77-0465	100mg	
	77-6100	50mg	
	77-6580	50mg	
	77-7015	50mg	
	77-7030	50mg	
77-0218	4,4'-Bis(t-butyl)-2,2'-bipyridine]bis[5-methyl-2-(4-methyl-2-pyridinyl-kN)phenyl-kC]iridium hexafluorophosphate, 95% (1607469-49-7)	50mg	See page 17
77-0285	[4,4'-Di-t-butyl-2,2'-bipyridine][bis[5-(t-butyl)-2-[4-(t-butyl)-2-pyridinyl-kN]phenyl-kC]iridium(III) hexafluorophosphate, 95% (808142-80-5)	50mg	See page 19
77-0410	(4,4'-Di-t-butyl-2,2'-bipyridine)bis[2-(2-pyridinyl-kN)phenyl-kC]iridium(III) hexafluorophosphate, 99% (676525-77-2)	100mg	See page 20
77-0425	(4,4'-Di-t-butyl-2,2'-bipyridine)bis[3,5-difluoro-2-[5-trifluoromethyl-2-pyridinyl-kN]phenyl-kC]iridium(III) hexafluorophosphate, 99% (870987-63-6)	50mg	See page 19
77-0453	(2,2'-Bipyridine)bis[3,5-difluoro-2-[5-trifluoromethyl-2-pyridinyl-kN]phenyl-kC]iridium(III) hexafluorophosphate, 99% (1092775-62-6)	50mg	See page 15
77-0465	(2,2'-Bipyridine)bis[2-pyridinyl-kN]phenyl-kC]iridium(III) hexafluorophosphate, 99% (106294-60-4)	100mg	See page 16
77-6100	Tris[5-fluoro-2-(2-pyridinyl-kN)phenyl-kC]iridium(III), 95% (370878-69-6)	50mg	See page 22
77-6580	Tris[(2-(2-pyridinyl-kN)-5-(trifluoromethyl)phenyl-kC]iridium(III), 95% (500295-52-3)	50mg	See page 23
77-7015	Tris(2-phenylpyridinato-C2,N)iridium(III), 95% (94928-86-8)	50mg	See page 22
77-7030	Tris[2-(2,4-difluorophenyl)pyridine]iridium(III), 95% (387859-70-3)	50mg	See page 21

**KITS - Iridium Photocatalyst Kit 2**

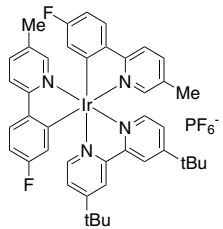
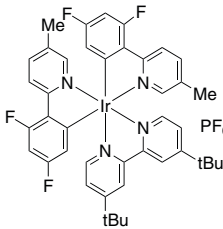
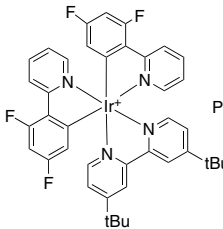
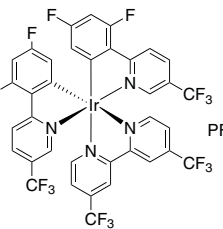
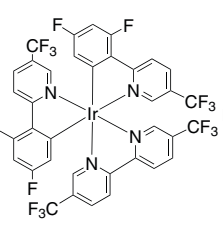
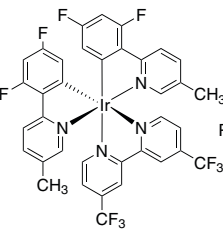
96-7790

**Iridium Photocatalyst Kit 2**

Components also available for individual sale.

Contains the following:

**NEW**

 <p>77-0320 50mg</p>	 <p>77-0330 100mg</p>	 <p>77-0350 100mg</p>
 <p>77-0360 50mg</p>	 <p>77-0370 50mg</p>	 <p>77-0380 50mg</p>
<p>77-0320</p> <p>[4,4'-Bis(1,1-dimethylethyl)-2,2'-bipyridine-κN,κN] bis[5-fluoro-2-(5-methyl-2-pyridinyl)phenyl-κC]iridium hexafluorophosphate, 98% (808142-88-3)</p>	<p>77-0330</p> <p>[4,4'-Bis(1,1-dimethylethyl)-2,2'-bipyridine-κN,κN] bis[3,5-difluoro-2-(5-methyl-2-pyridinyl)phenyl] iridium hexafluorophosphate, 98% (1335047-34-1)</p>	<p>77-0350</p> <p>[4,4'-Bis(1,1-dimethylethyl)-2,2'-bipyridine-κN,κN] bis[3,5-difluoro-2-(2-pyridinyl-κN)phenyl-κC]iridium hexafluorophosphate, 97% (1072067-44-7)</p>
<p>77-0360</p> <p>4,4'-Bis(trifluoromethyl)-2,2'-bipyridinebis[3,5-difluoro-2-[5-(trifluoromethyl)-2-pyridinyl]phenyl] iridium(III) hexafluorophosphate (2030437-90-0)</p>	<p>77-0370</p> <p>[5,5'-Bis(trifluoromethyl)-2,2'-bipyridine-κN,κN]bis[3,5-difluoro-2-[5-(trifluoromethyl)-2-pyridinyl-κN]phenyl]iridium hexafluorophosphate, 98% (1973375-72-2)</p>	<p>77-0380</p> <p>4,4'-Bis(trifluoromethyl)-2,2'-bipyridinebis[3,5-difluoro-2-[5-methyl-2-pyridinyl]phenyl] iridium(III) hexafluorophosphate</p>

**KITS - Iridium Photocatalyst Master Kit**

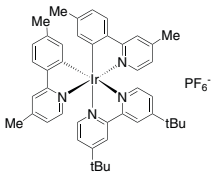
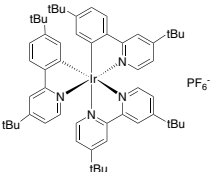
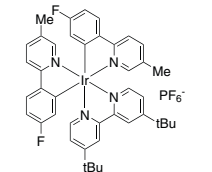
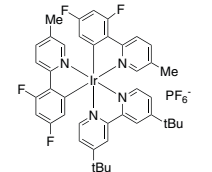
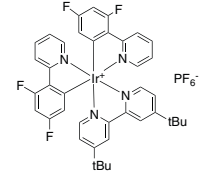
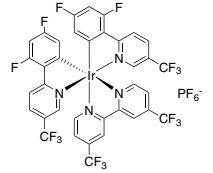
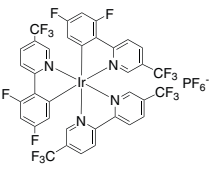
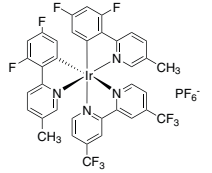
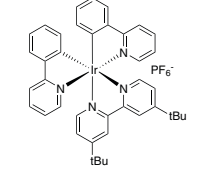
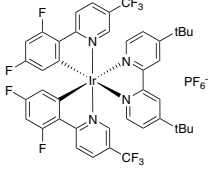
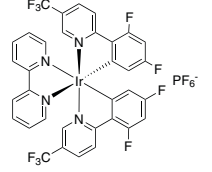
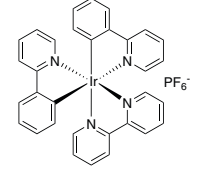
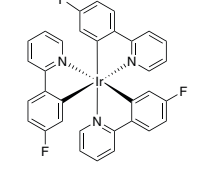
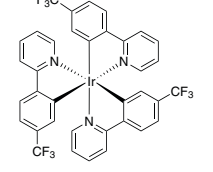
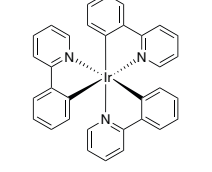
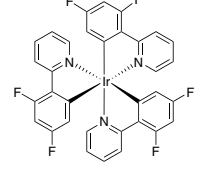
96-7795

**Iridium Photocatalyst Master Kit**

Components also available for individual sale.

**NEW**

Contains the following:

 <p>77-0218 50mg</p>	 <p>77-0285 50mg</p>	 <p>77-0320 50mg</p>	 <p>77-0330 100mg</p>
 <p>77-0350 100mg</p>	 <p>77-0360 50mg</p>	 <p>77-0370 50mg</p>	 <p>77-0380 50mg</p>
 <p>77-0410 100mg</p>	 <p>77-0425 50mg</p>	 <p>77-0453 50mg</p>	 <p>77-0465 100mg</p>
 <p>77-6100 50mg</p>	 <p>77-6580 50mg</p>	 <p>77-7015 50mg</p>	 <p>77-7030 50mg</p>

**KITS - Iridium Photocatalyst Master Kit**

96-7795 (continued)	Iridium Photocatalyst Master Kit		
77-0218	4,4'-Bis(t-butyl-2,2'-bipyridine)bis[5-methyl-2-(4-methyl-2-pyridinyl-kN)phenyl-kC]iridium hexafluorophosphate, 95% (1607469-49-7)	50mg	See page 17
77-0285	[4,4'-Di-t-butyl-2,2'-bipyridine]bis[5-(t-butyl)-2-[4-(t-butyl)-2-pyridinyl-kN]phenyl-kC]iridium(III) hexafluorophosphate, 95% (808142-80-5)	50mg	See page 19
77-0320	[4,4'-Bis(1,1-dimethylethyl)-2,2'-bipyridine-kN,kN] bis[5-fluoro-2-(5-methyl-2-pyridinyl-kN)phenyl-kC]iridium hexafluorophosphate, 98% (808142-88-3)	50mg	See page 17
77-0330	[4,4'-Bis(1,1-dimethylethyl)-2,2'-bipyridine-kN,kN] bis[3,5-difluoro-2-(5-methyl-2-pyridinyl)phenyl] iridium hexafluorophosphate, 98% (1335047-34-1)	100mg	See page 17
77-0350	[4,4'-Bis(1,1-dimethylethyl)-2,2'-bipyridine-kN,kN] bis[3,5-difluoro-2-(2-pyridinyl-kN)phenyl-kC]iridium hexafluorophosphate, 97% (1072067-44-7)	100mg	See page 17
77-0360	4,4'-Bis(trifluoromethyl)-2,2'-bipyridinebis[3,5-difluoro-2-[5-trifluoromethyl-2-pyridinyl)phenyl] iridium(III) hexafluorophosphate (2030437-90-0)	50mg	See page 18
77-0370	[5,5'-Bis(trifluoromethyl)-2,2'-bipyridine-kN,kN]bis[3,5-difluoro-2-[5-(trifluoromethyl)-2-pyridinyl-kN]phenyl]iridium hexafluorophosphate, 98% (1973375-72-2)	50mg	See page 18
77-0380	4,4'-Bis(trifluoromethyl)-2,2'-bipyridinebis[3,5-difluoro-2-[5-methyl-2-pyridinyl)phenyl] iridium(III) hexafluorophosphate	50mg	See page 17
77-0410	(4,4'-Di-t-butyl-2,2'-bipyridine)bis[2-(2-pyridinyl-kN)phenyl-kC] iridium(III) hexafluorophosphate, 99% (676525-77-2)	100mg	See page 20
77-0425	(4,4'-Di-t-butyl-2,2'-bipyridine)bis[3,5-difluoro-2-[5-trifluoromethyl-2-pyridinyl-kN)phenyl-kC]iridium(III) hexafluorophosphate, 99% (870987-63-6)	50mg	See page 19
77-0453	(2,2'-Bipyridine)bis[3,5-difluoro-2-[5-trifluoromethyl-2-pyridinyl-kN)phenyl-kC]iridium(III) hexafluorophosphate, 99% (1092775-62-6)	50mg	See page 15
77-0465	(2,2'-Bipyridine)bis[2-pyridinyl-kN)phenyl-kC]iridium(III) hexafluorophosphate, 99% (106294-60-4)	100mg	See page 16
77-6100	Tris[5-fluoro-2-(2-pyridinyl-kN)phenyl-kC]iridium(III), 95% (370878-69-6)	50mg	See page 22
77-6580	Tris[(2-(2-pyridinyl-kN)-5-(trifluoromethyl)phenyl-kC] iridium(III), 95% (500295-52-3)	50mg	See page 23
77-7015	Tris(2-phenylpyridinato-C2,N)iridium(III), 95% (94928-86-8)	50mg	See page 22
77-7030	Tris[2-(2,4-difluorophenyl)pyridine]iridium(III), 95% (387859-70-3)	50mg	See page 21

**KITS - Ruthenium Photocatalyst Kit**

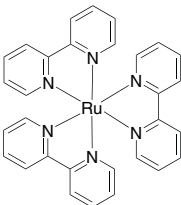
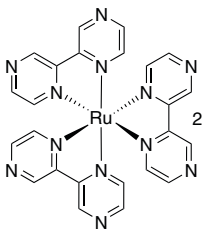
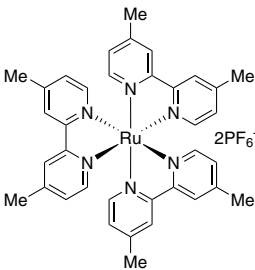
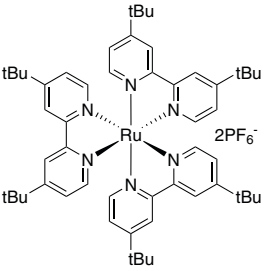
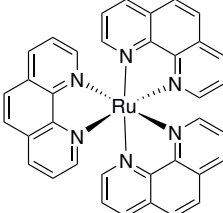
96-4450

**Ruthenium Photocatalyst Kit**

Components also available for individual sale.

Contains the following:

**NEW**

 <p>2Cl<sup>-</sup> 6H<sub>2</sub>O</p> <p>44-7900 250mg</p>	 <p>2PF<sub>6</sub><sup>-</sup></p> <p>44-7910 50mg</p>	 <p>2PF<sub>6</sub><sup>-</sup></p> <p>44-7930 50mg</p>	
 <p>2PF<sub>6</sub><sup>-</sup></p> <p>44-7940 50mg</p>	 <p>2PF<sub>6</sub><sup>-</sup></p> <p>44-7955 50mg</p>		
<p>44-7900</p>	<p>Tris(2,2'-bipyridyl)ruthenium(II) chloride hexahydrate, min. 98% (50525-27-4)</p>	<p>250mg</p>	<p>See page 28</p>
<p>44-7910</p>	<p>Tris(2,2'-bipyrazine)ruthenium(II) hexafluorophosphate, 95% (80907-56-8)</p>	<p>50mg</p>	<p>See page 27</p>
<p>44-7930</p>	<p>Tris(4,4'-dimethyl-2,2'-bipyridine)ruthenium(II) hexafluorophosphate, 95%, DMBPY (83605-44-1)</p>	<p>50mg</p>	<p>See page 29</p>
<p>44-7940</p>	<p>Tris[4,4'-bis(t-butyl)-2,2'-bipyridine]ruthenium(II) hexafluorophosphate, 95% (75777-87-6)</p>	<p>50mg</p>	<p>See page 28</p>
<p>44-7955</p>	<p>Tris(1,10-phenanthroline)ruthenium(II) hexafluorophosphate, 95% (60804-75-3)</p>	<p>50mg</p>	<p>See page 29</p>

**PHOTOCHEMICAL EQUIPMENT**

**98-7500** **EvoluChem™ PhotoRedOx Box**  
 Note: Sold in collaboration with HepatoChem

1 pc

**NEW**



The EvoluChem™ PhotoRedOx Box device is designed to facilitate photochemical experiments. This device is compatible with most vial formats (see related Photochemistry holders: 98-7600, 98-7650 or 98-7700). Its compact design allows for use with any stirring plate. A built-in fan keeps the reaction conditions at room temperature.

**Features**

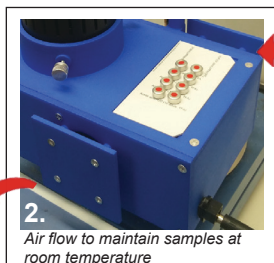
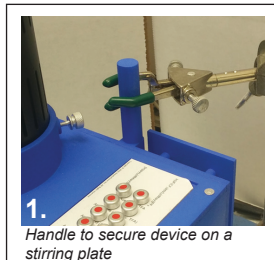
- Light source (See 98-7800)
- Photochemistry chamber to evenly distribute light
- Flexible vial formats
- Magnetic stirring on standard stirring plate
- Cooling by fan to maintain experiment at room temperature
- Pre-designed array of catalysts and reagents available

**Benefits**

- Easy set-up on a standard stirring plate
- Performs up to 32 reaction conditions simultaneously
- Individually sealed vials enable flexible study design
- Save your substrate using low scale reaction conditions
- Save time on optimization

**Easy set-up and compact design (see images on left)**

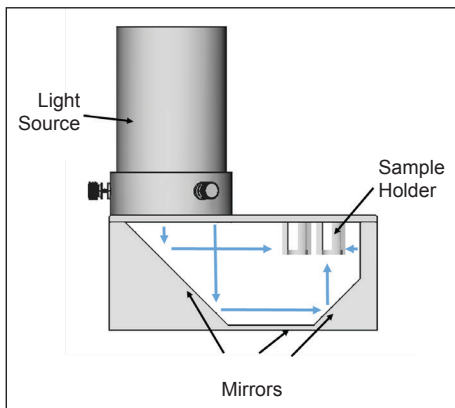
1. Handle to secure device on a stirring plate
2. Air flow to maintain samples at room temperature



**Unique Geometry to focus light on samples**  
 EvoluChem™ PhotoRedOx Box is equipped with several mirrors that direct and distribute the light toward the samples. The geometry of the box enables parallel reaction with homogeneous light exposure.

**Better Heat Management**

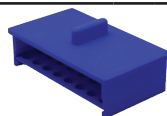
The position of the light source on the side of the samples reduces the amount of heat directed to the samples. The embedded fan eliminates any remaining heat.



**98-7600** **EvoluChem™ PhotoRedOx Box Photochemistry Holder**  
 32 x 0.3ml vials  
 Note: Sold in collaboration with HepatoChem

1 pc

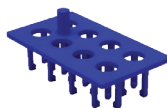
**NEW**



**98-7650** **EvoluChem™ PhotoRedOx Box Photochemistry Holder**  
 8 x 2ml vials  
 Note: Sold in collaboration with HepatoChem

1 pc

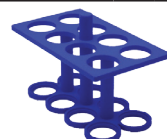
**NEW**



**98-7700** **EvoluChem™ PhotoRedOx Box Photochemistry Holder**  
 8 x 8ml vials  
 Note: Sold in collaboration with HepatoChem

1 pc

**NEW**



**PHOTOCHEMICAL EQUIPMENT**

98-7800

**EvoluChem™ PhotoRedOx Box Light Source**

**Wavelength 450nm, Electric Power 18W**

Note: Sold in collaboration with HepatoChem

1 pc

**NEW**



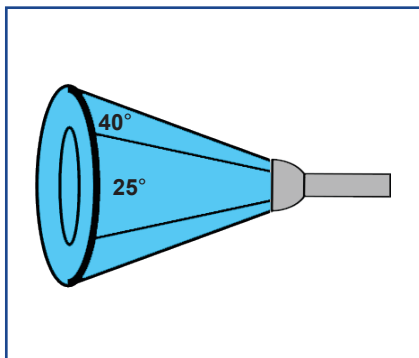
The EvoluChem™ light source is designed specifically for photocatalytic chemistry applications. It fits the EvoluChem™ PhotoRedOx Box (98-7500) and is designed to irradiate all samples with maximum efficiency. The LED chips are selected for specific wavelengths.

**General Specifications**

Power Consumption	18W
Input Voltage	100-240 VAC
Beam Angle	25°
Wavelength Options	450nm
LED	Cree XPE

**Light Power vs. Irradiance**

Although the total power of LED light is important, it is essential to estimate the amount of light that actually goes on the sample. If the light is spread over a large area the density of light (irradiance) on sample will be little. Therefore we designed the EvoluChem™ LEDs to focus the light toward the samples at a 25° angle.



*Focused Light Beam*



*Directly compatible with PhotoRedOx Box 98-7500*

## PHOTOCATALYST KITS - Compatible with PhotoRedOx Box

96-7510

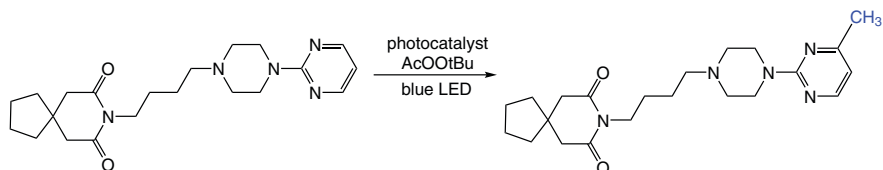
EvoluChem™ Photochemical Methylation Array Kit

1 kit

NEW

Note: Sold in collaboration with HepatoChem

This kit and the PhotoRedOx Box (98-7500) work together seamlessly.



Reference: *Chem. Soc. Rev.*, **2016**, *45*, 546-576

**Kit Protocol:**

The typical protocol is performed in a 0.05 Mol/l concentration reaction condition using a substrate solution of four different solvents. Each sealed reaction vial contains 0.1 μmol of photocatalyst and 12.5 μmol of *tert*-butyl peracetate. Based on the concentration of the substrate stock solution and the volume added, the following reaction stoichiometry can be achieved with the standard photomethylation kit.

	77-0425	77-0410
50/50 Acetonitrile/TFA	5 equiv. <i>tert</i> -butyl peracetic acid	
Acetonitrile (10 equiv. TFA)		
Acetic acid (10 equiv. TFA)		
Acetic acid/H <sub>2</sub> O (10 equiv. TFA)		

**Kit contents:**

Description	Quantity	Amount
(4,4'-Di- <i>t</i> -butyl-2,2'-bipyridine)bis[3,5-difluoro-2-[5-(trifluoromethyl)-2-pyridinyl- <i>kN</i> ]phenyl- <i>kC</i> ]iridium(III) hexafluorophosphate, 99% <b>Item # 77-0425</b>	8 vials	0.1 μmol/12.5 μmol
(4,4'-Di- <i>t</i> -butyl-2,2'-bipyridine)bis[2-(2-pyridinyl- <i>kN</i> )phenyl- <i>kC</i> ]iridium(III) hexafluorophosphate, 99% <b>Item # 77-0410</b>	8 vials	0.1 μmol/12.5 μmol
50/50 Acetonitrile/ trifluoroacetic acid	1 vial	1 ml
Acetonitrile (10 equiv. trifluoroacetic acid*)	1 vial	1 ml
Acetic acid (10 equiv. trifluoroacetic acid*)	1 vial	1 ml
Acetic acid/water (10 equiv. trifluoroacetic acid*)	1 vial	1 ml
Substrate stock vial 1	1 vial	--
Substrate stock vial 2	1 vial	--
Substrate stock vial 3	1 vial	--
Substrate stock vial 4	1 vial	--



## PHOTOCATALYST KITS - Compatible with PhotoRedOx Box

96-7560

EvoluChem™ Photocatalytic Alkylation Kit

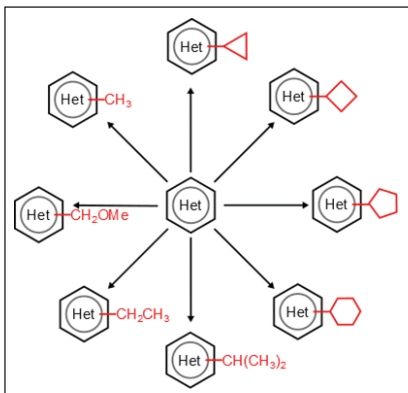
1 kit

NEW

Note: Sold in collaboration with HepatoChem

## Product Overview:

The trifluoroborate alkylation reaction (Minisci reaction)<sup>1</sup> is a powerful late stage functionalization tool. Our kit allows convenient, one-step production of eight different analogues of a lead compound in mg quantities. Each reaction vial contains 75 µmol of trifluoroborate alkylation reagent (pre-weighed) and a stirring bar to react with 50 µmol of substrate. C-H functionalization will primarily occur on electron-deficient heteroarenes at one or several positions.



## Kit Contents (16 reaction vials total):

- 2 reaction vials of  $\text{BF}_3\text{K}$  reagents (75 µmol)
- 2 reaction vials of  $\text{K}_2\text{S}_2\text{O}_8$  (100 µmol)
- 2 vials of photocatalysts
- 2 vials of TFA

## Kit Protocol:

For each kit, 4mL of a 0.1 M solution of substrate (400 µmol total) in DMSO is prepared with 8.98 mg photocatalyst  $\text{Ir}(\text{dF-CF}_3\text{-ppy})_2(\text{dtbbpy})$  (77-0425) (8 µmol, 2 mol%) and trifluoroacetic acid (153 µL, 5 equiv) included. The solution is sparged with nitrogen. Each vial contains 27.0mg  $\text{K}_2\text{S}_2\text{O}_8$  (100 µmol, 2 equiv.) and 1.5 equiv.  $\text{BF}_3\text{K}$  reagent (75 µmol) in 2ml vials equipped with a stir bar and Teflon septa. Alternatively for methylation, vials contain

39.9 µL of tert-butyl peracetate (TBPA). Vials are prepared under argon. 500µL of substrate solution is added via syringe and the vial is placed in PhotoRedOx Box (98-7500) equipped with light source. Reaction is stirred for 2-24 hr.

## Photocatalytic Alkylation Reagents (2 Vials of each)

	cyclopropyl	cyclobutyl	cyclopentyl	cyclohexyl	ethyl	isopropyl	methoxy methyl	t-butyl peracetate
MW (g/mol)	147.98	162.00	176.03	190.06	135.97	149.99	151.97	132.16
CAS #	1065010-87-8	1065010-88-9	1040745-70-7	446065-11-8	44248-07-9	1041642-13-0	910251-11-5	107-71-1

## References:

1. *Chem. Sci.*, **2017**, 8 (39), 3512-3522
2. *Chem. Soc. Rev.*, **2016**, 45, 546-576

## PHOTOCATALYST KITS - Compatible with PhotoRedOx Box

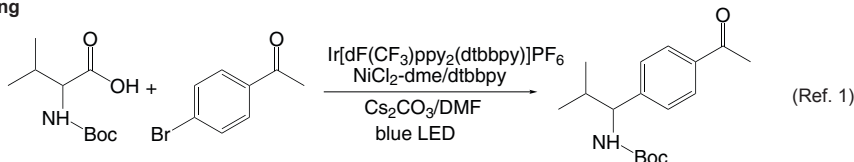
## Iridium/Nickel Photoredox Kits

Photoredox chemistry has been reported in literature using a wide range of catalysts and reagents. However, often these reactions are highly substrate, solvent and base specific. In order to facilitate the screening of common photochemistry reactions, HepatoChem has released a series of kits combining common Iridium, Nickel, ligand and base combinations to achieve successful cross-coupling transformations.

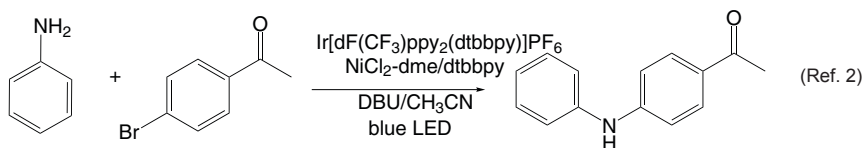
## Ir/Ni catalysis versatility

Depending on the ligand, base and solvent, the Ir/Ni catalytic systems can perform different cross-coupling reaction.

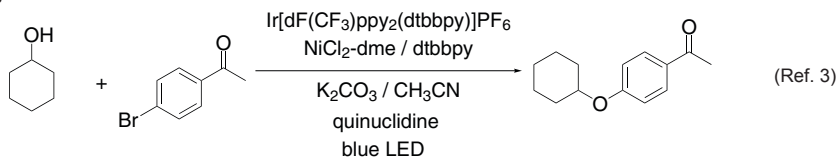
## C-C Coupling



## C-N Coupling



## C-O Coupling



## Several Kits Available

## Standard Protocol:

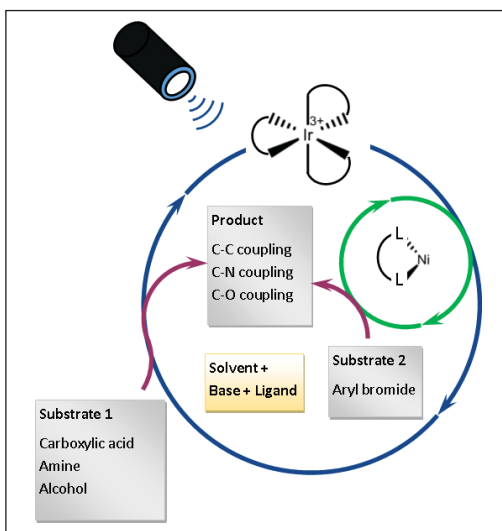
5  $\mu\text{mol}$  of substrates in 100  $\mu\text{l}$  solvent with Ir catalyst (2 mol %),  $\text{NiCl}_2 \cdot \text{dme}$  (10 mol %), ligand (10 mol %), and 3 equivalent of base.

## Features:

- 0.3ml vial with crimp cap and stirring bar
- Specifically designed for photchemistry device
- Pre-weighed reagents and catalysts
- Temperature maintained at RT
- Pre-designed or custom arrays available
- Reagents are packaged under inert atmosphere

## References

1. *Science* **2014**, *345*, 437-440
2. *Angew. Chemie*, **2016**, *55*, 13219-13223
3. *Nature* **2015**, *524*, 330-334



**PHOTOCATALYST KITS - Compatible with PhotoRedOx Box****Iridium/Nickel Photoredox Kits (continued)****Results summary:**

Selection of base and solvent is important to find the condition for appropriate coupling (5µmol per reaction/100µL scale)

Reaction Type	Substrates	Solvent	Base			
			Cs <sub>2</sub> CO <sub>3</sub>	K <sub>3</sub> PO <sub>4</sub>	DABCO	DBU
C-C coupling through decarboxylation	Boc-Val 4-bromoacetophenone	DMF	✓	✓		
C-N coupling (secondary amines)	Pyrolidine 4-bromoacetophenone	DMA			✓	
C-N coupling (aromatic amine/secondary amine)	Indoline 4-bromoacetophenone	DMA		✓		
C-N coupling (aromatic amine)	Aniline 4-bromoacetophenone	ACN			✓	✓

**96-7520 EvluChem™ Iridium/Nickel PhotoRedOx Base and Solvent Screening Kit 1**

1 kit

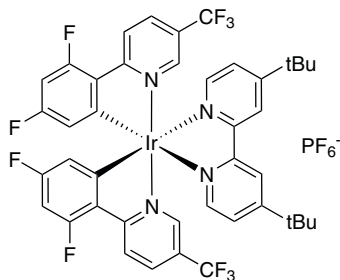
**NEW**

Note: Sold in collaboration with HepatoChem

**Kit Contents:**

This kit contains 77-0425 (2 mol%), Ni/Ligand (10 mol%) and base (3 eq)

	Cs <sub>2</sub> CO <sub>3</sub>	K <sub>3</sub> PO <sub>4</sub>	K <sub>2</sub> HPO <sub>4</sub>	KOH	Li <sub>2</sub> CO <sub>3</sub>	K <sub>2</sub> CO <sub>3</sub>	DABCO	DBU
Solvent A	2 sets of 8 conditions with 8 different bases per kit (16 total vials)							
	5 µmol of substrates in 100 µl solvent							
Solvent B	77-0425 (2 mol%), Ni/Ligand (10 mol%) and base (3 eq)							

**Iridium Catalyst: 77-0425****Suggested Solvents (not included)**

1. ACN
2. DMF
3. DMA
4. DMSO

**PHOTOCATALYST KITS - Compatible with PhotoRedOx Box**

96-7530

**EvoluChem™ Iridium/Nickel PhotoRedOx Base and Ligand Screening Kit 2**

1 kit

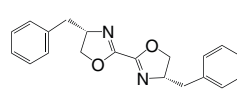
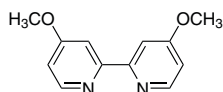
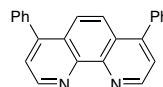
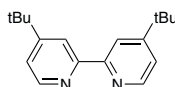
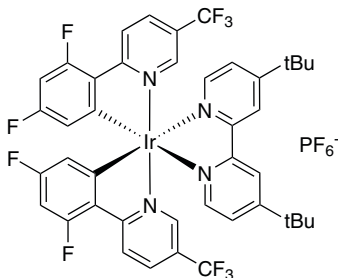
**NEW**

Note: Sold in collaboration with HepatoChem

**Kit Contents:**

This kit contains 77-0425 (2 mol%), Ni/Ligand (10 mol%) and base (3 eq)

	$\text{Cs}_2\text{CO}_3$	$\text{K}_3\text{PO}_4$	$\text{K}_2\text{HPO}_4$	$\text{K}_2\text{CO}_3$
dtbbpy	2 sets of 16 conditions with 4 bases and 4 ligands per kit (32 total vials) 5 $\mu\text{mol}$ of substrates in 100 $\mu\text{l}$ solvent 77-0425 (2 mol%), Ni/Ligand (10 mol%) and base (3 eq)			
bphen				
(MeO) <sub>2</sub> bpy				
biox				



96-7540

**EvoluChem™ Iridium/Nickel PhotoRedOx Base and Ligand Screening Kit 3**

1 kit

**NEW**

Note: Sold in collaboration with HepatoChem

**Kit Contents:**

This kit contains 77-0425 (2 mol%), Ni/Ligand (10 mol%) and base (3 eq)

	$\text{Cs}_2\text{CO}_3$	$\text{K}_3\text{PO}_4$	$\text{K}_2\text{HPO}_4$	$\text{K}_2\text{CO}_3$	DABCO	DBU
dtbbpy	2 sets of 24 conditions with 6 bases and 4 ligands per kit (48 total vials) 5 $\mu\text{mol}$ of substrates in 100 $\mu\text{l}$ solvent 77-0425 (2 mol%), Ni/Ligand (10 mol%) and base (3 eq)					
bphen						
(MeO) <sub>2</sub> bpy						
biox						

See catalyst and ligand structures with 96-7530.

## PHOTOCATALYST KITS - Compatible with PhotoRedOx Box

96-7550

EvoluChem™ Iridium/Nickel PhotoRedOx Base and Iridium Catalyst Screening Kit

1 kit

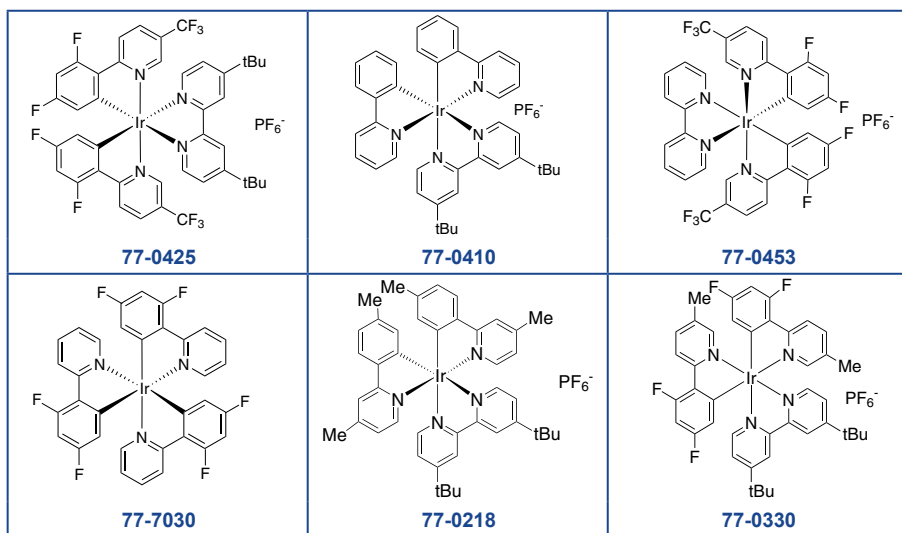
NEW

Note: Sold in collaboration with HepatoChem

## Kit Contents:

This kit contains Ir catalyst (2 mol%), Ni/Ligand (10 mol%) and base (3 eq)

	Cs <sub>2</sub> CO <sub>3</sub>	CsF	DBU
77-0425	2 sets of 18 conditions with 3 bases and 6 Ir catalysts per kit (36 total vials) 5 μmol of substrates in 100 μl solvent Ir catalyst (2 mol%), Ni/Ligand (10 mol%) and base (3 eq)		
77-0410			
77-0453			
77-7030			
77-0218			
77-0330			



96-7570

EvoluChem™ Iridium/Nickel PhotoRedOx Base and Solvent Screening Kit 2 (C-O coupling)

1 kit

NEW

Note: Sold in collaboration with HepatoChem

## Kit Contents:

This kit contains 2 sets of 8 reaction conditions per kit (16 total vials) with 77-0425 (1 mol%), Ni/Ligand and quinuclidine

Condition 1	Condition 2	Condition 3	Condition 4	Condition 5	Condition 6	Condition 7	Condition 8
Cs <sub>2</sub> CO <sub>3</sub> 1.5 eq.	K <sub>3</sub> PO <sub>4</sub> 1.5 eq.	K <sub>2</sub> CO <sub>3</sub> 1.5 eq.	K <sub>2</sub> CO <sub>3</sub> 1.5 eq.	K <sub>2</sub> CO <sub>3</sub> 1.5 eq.	DABCO 1.5 eq.	Quinuclidine 1.5 eq.	No Base Control
NiCl <sub>2</sub> -dme/ dtbbpy 5 mol%	NiCl <sub>2</sub> -dme/ dtbbpy 5 mol%	NiCl <sub>2</sub> -dme/ dtbbpy 5 mol%	NiCl <sub>2</sub> -dme/ dtbbpy 2.5 mol%	NiCl <sub>2</sub> -dme/ dtbbpy 1.25 mol%	NiCl <sub>2</sub> -dme/ dtbbpy 5 mol%	NiCl <sub>2</sub> -dme/ dtbbpy 5 mol%	NiCl <sub>2</sub> -dme/ dtbbpy 5 mol%
Quinuclidine 10 mol%							
77-0425 1 mol%							

The background of the entire image is a faded, light gray periodic table of elements. The elements are arranged in their standard grid format, with their symbols, atomic numbers, and names visible. The text is overlaid on this background.

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PHOTOCATALYST

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